NOVEL SYNTHESES OF THE CARBAPENEM KEY INTERMEDIATES, (3R,4R)-4-ACETOXY-3-[(R)-1-(t-BUTYLDIMETHYLSILYLOXY)ETHYL]-2-AZETIDINONE AND (3S,4R)-3-[(R)-1-(t-BUTYLDIMETHYLSILYLOXY)ETHYL]-4-CARBOXYMETHYL-2-AZETIDINONE, FROM (S)-ETHYL LACTATE¹⁾

Yoshio Ito,^{a)} Yuko Kobayashi,^{a)} Takeo Kawabata,^{a)} Mitsuru Takase,^{b)} and Shiro Terashimaa)*

Sagami Chemical Research Center, Nishi-Ohnuma, Sagamihara, Kanagawa 229, Japana) Technical Research Laboratory, Nihongi Plant, Nippon Soda Co. Ltd., Fujisawa, Nakago-Mura, Nakakubiki-Gun, Niigata 949-23, Japanh)

(Received in Japan 4 April 1989)

Abstract: Two types of the carbapenem hey intermediates (4 and 6) have been efficiently synthesized from inexpensive (S)-ethyl lactate (7). Thus, (S)-2-benzyloxypropanal readily obtainable from 7 was condensed with di*p*-anisylmethylamine to give the chiral imine. The [2+2]-cycloaddition *reaction of dihetene with the imine underwent in a highly stereoselective manner, yielding the desired 3,4-trans-3-acetyl-6-lactam (13a) as a major product (diastereoselectivity 7-1O:l). This was elaborated to 4 and 6 in 9 and 6 steps, respectively.*

The carbapenem antibiotics represented by thienamycin (l), have been the focus of recent synthetic attention because these novel compounds exhibit prominent antibacterial activities, broad spectra, and high resistance to bacterial 6-lactamases together with unique framework, specifically the 1-carbapen-2-em nucleus, and $6-(1-hydroxvethv)$ substituent in place of the traditional amide functionality.²⁾ In several syntheses of these novel antibiotics so far reported,³⁾ the latter title compound, $(3S,4R)$ -3- $[(R)-1-(t-butyldimethylsilyloxy)ethyl]-4-carboxymethyl-2-aze$ tidinone (4), or its equivalents have ingeniously been employed as key synthetic intermediates⁴⁾ in the synthesis of **1** and various novel synthetic routes to these important compounds have hitherto been explored.⁵⁾ The former title compound, $(3R, 4R)$ -4-acetoxy-3-[(R)-1-(tbutyldimethylsilyloxy)ethyl]-2-azetidinone (6) ,⁶⁾ holds more pivotal position than 4 as one of the most versatile carbapenem key intermediates since various types of carbon chains required to construct the five-membered ring fused with β -lactam, can be readily introduced into 6 by substituting its acetoxy group with nucleophiles.^{5b,c,7)} The same synthetic strategy has recently

been applied for producing the key intermediate $(5)^{8}$ of 1 β -methylcarbapenem such as 2^{9} and $3¹⁰$ chemically and metabolically more stable carbapenems exhibiting excellent antibacterial activities. It has also been uncovered that various penem compounds can be elaborated from 6 by substituting its acetoxy group with thiol derivatives.¹¹⁾ As the usefulness of 6 as a raw material in industries as well as in laboratories turns out to be apparent, a number of synthetic methods of 6 have hitherto been explored by employing various chiral compounds as starting materials. $5f,12$

Recently, we have also succeeded in exploring novel synthetic routes to 4 and 6 in which the stereoselective addition of diketene to the optically active imine **(10)** readily obtainable from commercially available inexpensive (S) -ethyl lactate (7) , plays a key role to construct the chiral 3.4 -trans-3-acetyl- β -lactam **(13)** bearing the desired absolute stereochemistry.¹⁾

It was recently disclosed that the addition reactions of diketene with imines derived from aromatic aldehydes¹³⁾ or alkyl glyoxylates,¹⁴⁾ can proceed in a stereoselective manner to afford $3,4-trans-3-acceptl$ - β -lactams. However, this novel β -lactam formation being formally a [2+2]cycloaddition reaction of acelylketene presumably produced from diketene, has never been examined with the imine prepared from an optically active aliphatic aldehyde carrying a chiral center at the α -position.¹⁵⁾ We have now found that the absolute stereochemistry of 3,4-trans-3acetyl-ß-lactam can be effectively controlled by the adjacent chiral center, and that the highly optically active 3,4-trans-3-acetyl-9-lactam **(MaI can** be readily elaborated to 4 and 6. This report concerns with full details of the novel syntheses of 4 and 6.

The $[2+2]$ -Cycloaddition Reaction of Diketene with the Chiral Imine (10) Prepared from (S)-**Ethyl Lactate**

As shown in Scheme **1, the** explored synthetic scheme commences with protection of the hydroxy group of 7. Thus, after conversion of 7 into the corresponding benzyl ether **@a)** by treating with 0-benzyl trichloroacetimidate in the presence of a catalytic amount of trifluoromethanesulfonic acid,^{16,17)} the ester group of 8a was effectively reduced with diisobutylaluminium hydride to afford (S) -2-benzyloxypropanal $(9a)$. Since studies on the $[2+2]$ -cyclo-

Scheme 1

a) CClsC(NH)OBn-TfOH, 69%; TBDMSCI-imidazole, 96%; DHP-PPTS, 72%; MEMCI-(Me2CH)2NEt, 59%; EtOCHCH2-TsOH, 91% b) DIBAL in ether, -78 °C, 82% (9a); 76% (9b); 81% (9c); 78% (9d); 82% (9e) c) DAM-NH₂-MgSO₄ in toluene, see text for the yield d) pyrrolidine, 95% e) BnCl-NaH, 87% or BnCI-NaOH- (C8H17)3MeNCl, 92% f) NaAl(OCH2CH2OMe)2H2 in toluene, 88%

addition reaction disclosed that the benzyl group is most favorable with respect to stereoselectivity of the addition reaction (vide infra), more economical synthetic method was sought which could afford 9a without use of expensive O-benzyl trichloroacetimidate. After numerous experimentations, it was finally found that $(S)-N$, N-tetramethylenelactamide (11) readily obtainable by heating 7 with pyrrolidine, could be derived to the corresponding benzyl ether (12) by treating with benzyl chloride and sodium hydride or with benzyl chloride and sodium hydroxide under usual phase transfer conditions. These methods of benzylation are clearly more inexpensive than the previous reaction. Reduction of the amide group of 12 was achieved with sodium bis(2-methoxyethoxy)aluminum hydride (Vitride®),¹⁸⁾ giving rise to 9a.

Treatment of $9a$ with di-p-anisylmethylamine (DAM-NH₂)¹⁹⁾ in the presence of magnesium sulfate as a dehydrating agent produced the chiral imine (10a), which was immediately sub-

Table 1. The [2+2]-cycloaddition reaction of the chiral imine (10) with diketene^{a)}

a) The reaction was performed at -35 °C for 3 days with $4\neg 5$ equivalents of diketene and 1 equivalent of imidazole when otherwise not mentioned. Combined yield of 13 and 14 calculated based on 9. c) Determined by the $1H$ -NMR spectrum of the mixture of 13 and 14 when otherwise not mentioned. d) 4-Methylimidazole was used instead of imidazole. e) Ratio of CH2Cl2 to t-BuOH was 10:1. f) The reaction was quenched after 1 day. g) The ratio was determined by the $1H\text{-}NMR$ spectrum after the mixture of 13 and 14 was derived to that of 16f and 17f. See text.

jected to the next [2+2]-cycloaddition reaction **(Scheme 2).** Taking into account the previous re- $_{\rm{sults,14,20}}$ DAM-NH2 was selected as an amine counterpart for the imine formation. Addition of diketene to **10a** which constitutes the key stereoselective reaction of our synthetic route to 4 and 6 was first examined in tetrahydrofuran (THF) in the presence of 1 equivalent of imidazole. As shown in **Table 1** (run 1), a mixture of the 3,4-trans-3-acetyl- β -lactams (13a and 14a) could be produced in 78% yield.²¹⁾ The stereoselectivity of the $[2+2]$ -cycloaddition reaction was determined as 8.0:1 by integrating the acetyl protons which appeared as two singlets in the 1 H-NMR spectrum of the mixture of **13a and 14a. The** major isomer **(13a)** isolated as an oil by separating the mixture of 13a and 14a with column chromatography, showed $[\alpha]_{D^{20}}$ -7.3° (CHCls). The enantiomeric excess of 13a was estimated to be 96%ee by the ¹H-NMR spectrum measured in the presence of the chiral shift reagent, $\left[\text{Eu(hfc)}\right]$ ²²⁾ Successful preparations of 4 and 6 (vide infra) obviously established the absolute stereochemistry of 13a. By repeating the same [2+2]cycloaddition reactions in various solvents (runs $2.4 - 8$), it appeared evident that dichloromethane (CH2Cl2) is the solvent of choice with respect to the chemical yield and stereoselectivity (run 2). The use of 4-methylimidazole in place of imidazole was ineffective for improving the stereoselectivity and chemical yield of the [2+21-cycloaddition reaction (run 3). This result distinctly differs from that previously observed for the synthesis of 5 by the [2+21-cycloaddition of ketene with a chiral imine. 20

In order to evaluate the effect of protective group on the stereoselectivity and chemical yield, the E2+21-cycloaddition reactions were further examined employing various types of the imines **(lob-e) which carry** different protective groups. **The** requisite aldehydes **(9b-e)** could be prepared by way of **8b-e** by sequential protection and reduction. The [2+2]-cycloaddition reactions were attempted in THF or CH₂Cl₂ with 10b-e similarly produced from 9b-e and DAM-NH₂. As summarized in **Table 1,** it appeared obvious that exchanges of the protective groups from benzyl to other groups resulted in the decrease of stereoselectivity in the reactions in THF (runs 9,ll and 14) while the high stereoselectivity similar to that observed for **10a** could be recorded in the reactions in $CH2Cl2$ (runs 10, 12 and 13).

Based on the extensive studies described above, we have succeeded in firmly establishing the structure of imine and the reaction conditions which can afford the best result of the 12+21-cycloaddition reaction. Thus, the reaction performed with diketene (5 equivalents) and **10a** in CH2C12 at -35OC for 3 days successfully produced a mixture of **13a** and **14a** in a ratio of 7.3:1 and in 91% combined yield (run 2). While **13a** and **14a** were separable by column chromatography, the diastereomers formed at this stage could be nicely separated by recrystallization after the stereoselective reduction of the 3-acetyl group to the $3-(R)$ -1-hydroxyethyll group *(vide infra)*.

It is quite ambiguous whether the true reactant of the $[2+2]$ -cycloaddition reaction is diketene, acetylketene, or 1-(acetoacetyl)-imidazole. However, since stereoselectivity and chemical yield of the reaction highly depend upon polarity of solvent, it may be reasonable to expect that the $[2+2]$ -cycloaddition reaction proceeds through the zwitter-ionic intermediate such as **15** as previously claimed.^{15b,23}) While the reaction mechanism is still unclear, it is of interest that the methyl and alkoxy groups of **16** can effectively control stereoselectivity of the reaction even though difference of their steric bulkiness is not large enough.

Determination of the Stereochemistry of the [2+2]-Cycloaddition Products

In the explored $[2+2]$ -cycloaddition reaction, the two diastereomeric 3,4-trans-3-acetyl- β lactams (18 and 14) were regularly produced. While the absolute stereochemistries of 13a and 14a could be firmly established by the successful transformations of 13a into 4 and 6 (vide infra), the absolute stereochemistries of other cycloaddition products $(13b - e$ and $14b - e$) were determined by chemical correlations.

Thus, the diastereomeric 3,4-trans-3-acetyl- β -lactams (13a and 14a) were first subjected to acetalization with trimethyl orthoformate in the presence of a catalytic amount of camphorsufonic acid, producing the dimethyl acetals (16a and 17a), respectively. Under the same conditions as applied to 13a and 14a, 1Sb and 14b were directly transformed to the dimethyl acetals (16f and 17f), respectively, with concomitant cleavage of the t -butyldimethylsilyloxy group. The diastereomeric mixture of 13c and 14c and that of 13e and 14e were similarly derived to the diastereomeric mixture of 16f and 17f without change of the diastereomeric ratios, On the other hand, benzylation of 16f and 17f under the phase transfer conditions similar to those applied for preparing 12 from 7, gave 16a and 17a, respectively. These products were identified with authentic 16a and 17a prepared from 13a and 144 respectively. Since the 2-methoxyethoxymethyl groups present in 13d and 14d could not be removed under weakly acidic conditions, acetalization of the mixture of 13d and 14d gave rise to a mixture of the dimethyl acetals (16d and 17d). These products were identified with authentic 16d and 17d, respectively, which were produced from 16f and 17f by treating with 2-methoxyethoxymethyl chloride and base.

Based on these chemical correlations, the absolute stereochemistries of 13a-e and 14a-e could be rigorously established as shown in Table 1 and Scheme 2.

Preparation of the Carbapenem Key Intermediate (6)

With a large quantity of 13a in hand, preparation of 6 was next examined as shown in Scheme 3. Reduction of the acetyl group of 13a with potassium tri-sec-butylborohydride in the presence of potassium iodide²⁴⁾ underwent highly stereoselectively to give a mixture of the two epimeric alcohols (18 and 19, 18:19=12:1) with 4% recovery of 13a. On the other hand, complete reduction of 13a was effected with potassium triethylborohydride,²⁵) yielding a mixture of 18 and 19 in the same stereoselectivity. The two epimeric alcohols (18 and 19) separated showed $[\alpha]_{\text{D}}^{26}$ -5.5° (CHCl3) and $[\alpha]_{\text{D}}^{26}$ +13.8° (CHCl3), respectively. The major alcohol (18) could be readily isolated in a pure state in 76% yield by a single recrystallization of the reduction products from isopropanol. It was also found that when a crude diastereomeric mixture of 13a and 14a produced under the best conditions of the [2+21-cycloaddition reaction was reduced with potassium triethylborohydride without separation and the crude mixture of reduction products was immediately recrystallized from isopropanol, an almost pure sample of 18 could be obtained in 58% yield based on 9a (vide *supra*). This result may be quite useful for a large scale preparation of 6. The undesired epimer (19) could be converted to 18 by the Mitsunobu reaction in an excellent combined yield.²⁶⁾

Oxidative removal of the di-p-anisylmethyl (DAM) group was first attempted at the stage of 18. It was effected by employing cerium(IV) ammonium nitrate $(CAN)^{27}$ to afford the N-unprotected β -lactam (20), $[\alpha]_D^{2\delta} + 61.5^{\circ}$ (CHCls). After protection of the hydroxy group of 20 in a form of t-butyldimethylsilyl ether, the produced benzyl ether (21), $[\alpha]_{D^{25}} + 32.5^{\circ}$ (CHCls), was subjected to hydrogenolysis, giving the alcohol (22) , α] n^{25} -11.5° (CHCls). Since direct transformation of 22 to 6 by oxidative cleavage of the 1,2-amido alcohol with sodium periodate in the presence of sodium acetate turned out to be fruitless, 28 the following two step procedure was examined. Thus, oxidation of 22 with chromium trioxide gave the ketone (23), $[\alpha]_{D^{25}}$ -14.3° (CHCls). The same ketone (23) could be also obtained by oxidation of 22 with a combination of N-chlorosuccinimide, dimethylsulfide, and triethylamine.29) The latter oxidation method is

 $25: R^1=H, R^2=TBDMS$

a) KBsec-BusH-KI in THF, 0 °C, 92% (18:19=12:1) or KBEtsH in THF, -78 °C, 98% (18:19=12:1) b) i) EtOzCNNCOzEt-PPhs-HCOOH, ii) &CO3 in MeOH, 99% (2 Steps) c) CAN in aq. MeCN, -10 $^{\circ}$ C, 93% d) TBDMSCl-imidazole in DMF, 97% e) H2-Pd/C in AcOEt, 100% f) CrOs in pyridine, 93% or NCS-MezS-EtsN in toluene, 95% g) TBDMSCl-DMAP in DMF, 95% h) Hz-Pd/C in AcOEt, 98% i) NCS-MezS-EtsN in toluene, 98% j) Na2S2Os-Na2HPO4 in aq acetone, 89% k) MCPBA in AcOEt, 93% or perphthalic acid in AcOEt, 81%

anticipated to be more promising in a large scale preparation since poisonous chromium trioxide is not utilized.³⁰⁾

Since CAN is fairly expensive and the oxidation with it sometimes makes trouble during work-up in a large scale reaction due to fine precipitates of the resulting cerium(II1) compounds, oxidation with sodium peroxodisulfate was next examined as an alternative deprotection method of DAM group. Accordingly, the alcohol (18) was converted to the ketone (26), $[\alpha]_{D^{20}}$ +27.4° (CHCla), by way of the silyl ether (24), $[\alpha]_D^{28}+19.8^{\circ}$ (CHCla), and the alcohol (25), $[\alpha]_D^{28}$ -29.1" (CHCla), following the synthetic scheme similar to that employed for the preparation of 23 from 20. Oxidations of 18, 24, and 25 with sodium peroxydisulfate did not give the corresponding deprotected products in high yields due to partial decomposition of the starting materials and/or the deprotected products. However, it was finally found that treatment of 26 with sodium peroxydisulfate in the presence of disodium hydrogenphosphate as buffer cleanly effected removal of the DAM group, giving rise to 28. Taking into account the operational simplicity, the latter synthetic route to 28 is anticipated to be more practical than the former in which the DAM group has been removed at the stage of 18.

Oxidation of 23 with m-chloroperbenzoic acid, 31 cleanly produced 6, mp 108-109 °C and $[\alpha]_{D^{25}}$ +47.8° (CHCls). Monoperphthalic acid obtainable from phthalic anhydride and hydrogen peroxide was also usable for this Baeyer-Villiger oxidation. The carbapenem key intermediate (6) was identified by comparing its physical and spectral data with those reported. $12b,f,g$)

Preparation of Thienamycin Intermediate (4)

With completion of the efficient synthesis of 6 from 7 by way of the $[2+2]$ -cycloadduct (13a), we next examined elaboration of 13a to the advanced carbapenem key intermediate (4). Although 6 has been recognized as a useful precursor to produce 4 by nucleophilic substitution at the C_4 -position,⁶ the direct preparation of 4 from 13a seems to be advantageous since 13a has

a) TsCl in pyridine, 0 °C, 82% (27a), 85% (27b) b) NaI in acetone, reflux, 90% (28a), 91% (28b) c) DBU in toluene, 100 "C, 91% **@9a), 24%** (29b) (see text) d) CAN in aq MeCN, -10 °C, 74% e) BHs in THF, rt, 1h, then, H2O2-aq NaOH, 22% (see text); 9-BBN in THF-EtzO, rt, 3 h, then, H2O2-aq NaOH, $0 °C$, 77% f) RuCls $(2 \text{ mol}\%)$ -NaI04 in CC14-MeCN-H30, rt, 69%

the same carbon framework as that involved in 4.

As shown in Scheme 4, the alcohol (26) prepared from 13a in 3 steps was readily transformed to the corresponding *p*-toluenesulfonate $(27a)$, $[\alpha]_D^{\infty}$ -22.2° (CHCls). However, it turned out to be fruitless to directly eliminate the tosyloxy group to produce the olefin (29a) by treating 27a with bases such as 1,8-diazabicyclo[S.4.Olundec-7-ene (DBU), potassium t-butoxide, pyridine, etc. Accordingly, preparation of **29a** was examined by way of the iodide (28al. Thus, treatment of **27a with sodium** iodide gave rise to **28a** as a mixture of the two diastereomers. Ratio of the two diastereomers could be roughly estimated as $5:2$ by the 1 H-NMR spectrum of the mixture. This can be explained by epimerization of the initially formed iodide during the substitution reaction. Without separation of the diastereomers, treatment of 28a with DBU cleanly produced 29a, α ₁b²⁰ +52.5° (CHCl3). Oxidative removal of the DAM group was effected with CAN without a cleavage of the silyl ether, yielding $29b$, $\left[\alpha\right]p^{20} - 24.5^{\circ}$ (CHCls). On the other hand, the DAM group was removed at the stage of 18 and the deprotected β -lactam (20) was converted to the iodide **28b** by way of 22 by the same synthetic steps as employed for preparing 28a from 26. While 28b was subjected to the elimination reaction in a similar manner to that for 28a, a 24% of 29b was only obtained due to partial decomposition of28b and/or 29b.

Hydroboration of **29b with** borane followed by the usual oxidative workup gave a mixture of the desired primary alcohol (30) and its regioisomer [a mixture of 22 and its $4-[R]-1$ -hydroxyethyl]-isomer] in a ratio of 11:9 in 40 % combined yield. However, the use of 9-borabicyclo[3.3.1] nonane in place of borane readily produced 30 as a single product, $[\alpha]_{D}^{30}$ -22.3° (CHC13). Oxidation of 30 was achieved smoothly by the procedure reported by Sharpless, et $a l$, 32 to furnish 4, mp 150~154 °C (decomp.) and α lo²⁰+16.1° (CHCls). This was identified with the authentic sample^{4f)} by comparisons of their physical and spectral data.

As mentioned above, we have succeeded in exploring novel synthetic routes to two types of the carbapenem key intermediates $(4 \text{ and } 6)$ by featuring the $[2+2]$ -cycloaddition reaction of diketene with a chiral imine as a key stereoselective reaction. The synthetic scheme explored may be characterized by various notable aspects including high stereoselectivity in the β -lactam formation, use of commercially available inexpensive 7 as a starting material, high chemical yields of the overall processes, and ingenious utilization of all the carbon framework of the $[2+2]$ -cycloaddition product for preparing 4.

Acknowledgement

The authors are indebted to Dr. M. Sunagawa, Research Laboratories, Research and Development Division, Sumitomo Pharmaceuticals Co. Ltd., for stimulating discussions. The authors also thank Prof. M. Shibasaki, Faculty of Pharmaceutical Sciences, Hokkaido University, and Dr. T. Iimori, Sagami Chemical Research Center, for generously providing us with the spectral data of 4.

Experimental

General. All melting points were determined with a Yamato **MP-21** melting point apparatus and were uncorrected. Measurements of optical rotation were performed with a Horiba SEPA-200 automatic digital polarimeter. IR spectral measurements were carried out with a JASCO A-202 diffraction grating infrared spectrometer. 'H-NMR spectra were measured with a Hitachi R-90H (90 MHz) and a Bruker AM spectrometer (400 MHz). All signals were expressed as ppm downfield from tetramethylsilane used as an internal standard (6-value). Mass spec**tra were taken with a Hitachi RMU-6MG** mass spectrometer. Wakogel C-200 and C-300 were used as an adsorbent for column chromatography. Kieselgel 60F254 (Merck) was used for preparative TLC.

(S)-Ethyl 2-Benzyloxypropionate (8a). Trifluoromethanesulfonic acid (0.100 ml, 1.13 mmol) was added to a solution of 7 (3.50 g, 29.6 mmol) and O-benzyl trichloroacetimidate¹⁶⁾ (11.0 ml. 59.2 mmol) in cyclohexane-CH2Cl2 (7:1) (80 ml) at rt. After stirring for 20h, the mixture was diluted with H2O (20 ml) and hexane (80 ml), then stirred for additional 3h. The white precipitates resulted was removed by filtration. The aqueous layer of the filtrate was separated and the organic layer was washed with satd aq NaHCO₃ and satd aq NaCl, then dried over anhyd MgSO4. Filtration and concentration in vacuo gave an oily residue which was purified by column chromatography (SiO2, hexane-AcOEt 1:0~30:1), giving 8a as a colorless oil (4.27 g, 69%), [α]_p²⁰ -74.5° (c 2.94, CHCl₃). IR (neat): 3000, 1750, 1455, 1271, 1200, 1142, 1067, 1027, 740, 702 cm⁻¹. ¹H-NMR (CDCl3): 1.29 (3H, t, J=7.1 Hz, MeCH2), 1.43 (3H, d, J=6.8Hz, MeCH), 4.05 (1H, q, J=6.8 Hz, MeCH), 4.22 (2H, q, J=7.1Hz, MeCH₂), 4.43, 4.70 (2H, two d, J=each 11.7 Hz, PhCH2), 7.33 (5H, s, C6H5). MS m/e: 179 (M-Et)⁺, 102.

(S)-Ethyl 2-(t-Butyldimethylsilyloxy)propionate (8b). Imidazole (2.66 g, 39.1 mmol) and tbutyldimethylchlorosilane (4.85 g, 32.2 mmol) were added to a solution of 7 (3.12g, 26.4 mmol) in DMF (26 ml). After stirring at rt for 1h, the mixture was diluted with hexane (150 ml) and H2O (20 ml). The organic layer was separated, washed with H2O, then dried over anhyd MgSO4. Filtration and concentration in vacuo gave an oily residue which was purified by distillation (bp 88 °C, 20 mmHg), giving 8c as a colorless oil (5.90g, 96%), $[\alpha]_D^{20}$ -28.7° (c 2.36, CHCl3). IR (neat): 2950, 2870, 1755, 1462, 1255, 1142, 977, 835, 780, 660 cm⁻¹. ¹H-NMR (CDCls): 0.08, 0.10 (6H, two s, Me2Si), 0.91 (9H, s, Me3C), 1.35 (3H, t, J=7.0 Hz, MeCH2), 1.35 (3H, d, J=6.7 Hz, MeCH), 4.18 (2H, q, J=7.0 Hz, CH₂Me), 4.31 (1H, q, J=6.7 Hz, CHMe). MS m/e: 217(M-Me)⁺, 175 (M-Bu)⁺, 159, 147.

(S)-Ethyl 2-Tetrahydropyranyloxypropionate (8c). Dihydropyran (6.00 ml, 65.8 mmol) and pyridinium p-toluenesulfonate (1.16 g, 4.62 mmol) were added to a solution of 7 (5.18 g, 43.9 mmol) in CH2Cl2 (100 ml). After stirring at rt for 1.5 h, the mixture was diluted with ether (200 ml). The ethereal solution was washed with aq NaCl and dried over anhyd MgSO4. Filtration and concentration in vacuo gave an oily residue which was purified with distillation (bp 76-78 °C, 1 mmHg), giving 8c as a colorless oil (6.41 g, 72%). This sample (8c) consisted of an almost equal amount of the two diastereomers due to its tetrahydropyranyl group. IR (neat): 2950, 2880, 1750, 1445, 1200, 980, 870, 718 cm⁻¹. ¹H-NMR (CDCls): 1.27 (3H, t. J=7.1 Hz, MeCH2), 1.39. 1.44 (3H, two d, J=6.9 Hz, MeCH), 1.5~1.8 (6H, m, (CH2)3), 3.4~4.0 (2H, m, OCH2CH2), 4.19 (2H, q, J=7.1 Hz, MeCH2), 4.40 (1H, q, J=6.9 Hz, MeCH), 4.71 (1H, m, OCHO). MS m/e: 144 (M- $(CH2)3O$ ⁺, 129 (M-COOEt)⁺, 101, 85.

(S)-Ethyl 2-(2-Methoxyethoxymethoxy)propionate (8d), N,N-Diisopropylethylamine (9.4 ml, 54 mmol) and 2-methoxyethoxymethyl chloride (6.20 ml, 54.3 mmol) were added to a solution of 7 $(4.25 \text{ g}, 35.9 \text{ mmol})$ in CH2Cl2 (35 ml) at 0 °C. After stirring at rt for 12h, the mixture was diluted with H2O and CH2Cl2, and the aqueous phase was separated. The organic phase was washed successively with satd ag NaHCO3, 1M HCl and sat ag NaCl, then dried over anhyd MgSO4. Filtration and evaporation in vacuo gave an oily residue which was purified with distillation (bp 84 °C, 1 mmHg), giving 8d (4.37 g, 59 %), $[\alpha]n^{\infty}$ -66.7° (c 1.17, CHCl3). IR (neat): 3000, 2950, 2900, 1742, 1442, 1268, 1100, 1020, 845 cm⁻¹. ¹H-NMR (CDCl3): 1.28 (3H, t, J=7.0 Hz, MeCH2), 1.43 (3H,d, J=6.8 Hz, MeCH), 3.39 (3H, s, MeO), 3.47~3.59, 3.70~3.81 (4H, two m, OC2H4O), 4.19 (2H, q, J=7.0 Hz, CH2Me), 4.27 (1H, q, J=6.8 Hz, CHMe), 4.79 (2H, s, OCH2O).

(S)-Ethyl 2-(1-Ethoxyethoxy)propionate (8e). Ethyl vinyl ether (18.4 ml, 192 mmol) and p-toluenesulfonic acid (ca 1 mg) were added to a solution of 7 (2.50 g, 21.2 mmol) in ether (18 ml) at 0 °C. After stirring at rt for 1h, the mixture was diluted with satd aq NaHCOs (18 ml) and the aqueous phase was separated. The organic phase was washed with satd aq NaCl, then dried

over anhyd MgSO4. Filtration and concentration *in vacua* gave an oily residue which was purified with column chromatography, affording 86 as a colorless oil (3.66 g, 91%). IR (neat): 3000,2950,1753,1446,1375,1275,1177,1148,1083,1057,1027,967,660 cm-'. 'H-NMR (CDCls): 1.1~1.5 (12H, m, Mex4), 3.4~3.9 (2H, m, MeCH2OCH), 4.20 (2H, t, J=7.2 Hz, CH2OCO), 4.18, 4.33 $(1H, two q, J=each 5.9 Hz, OCHC), 4.78 (1H, q, J=5.5 Hz, OCHO).$

WI-243enzylosypropanal (9a). a) Preparation from 8a. A 1.0 M solution of diisobutylaluminium hydride in hexane $(2.70 \text{ ml}, 2.70 \text{ mmol})$ was added slowly to a solution of 8a $(0.373g, ...)$ 1.79 mmol) in ether at -78 °C. After stirring at the same temperature for 10 min, the mixture was diluted successively with MeOH (0.1 ml) and Hz0 (0.27 ml), then warmed up to rt. After stirring for additional lh, the resulting suspension was filtered through a pad of celite and the collected materials were washed with ether. The combined filtrates were dried over anhyd MgSO4 and concentrated *in vacua. The* concentration residue was purified with column chromatography (SiO₂, hexane-AcOEt 16:1~9:1) to give **9a** as a colorless oil (0.241 g, 82%). [α]²⁵ -66.8° ($l=1$, neat) $(lit, {}^{31})$ $[\alpha]_{p}^{20}$ -65.85° ($l=1$, neat)). IR (neat): 3470, 3058, 3000, 2950, 1740, 1500, 1456, 1380, 1210, 1100, 741, 702 cm⁻¹. ¹H-NMR (CDCls): 1.32 (3H, d, J=6.8 Hz, Me), 3.88 (1H, dq, $J=1.8, 6.8$ Hz, Me C_H), 4.62 (2H, s, Ph CH_2), 7.35 (5H, s, CsHs), 9.66 (1H, d, J=1.8 Hz, CHO). MS m/e: 181 (M+OH)+, 135 (M-CHO)+.

b) Preparation from 12. A 2.0 M solution of sodium bis(2-methoxyethoxy)aluminium hydride in toluene (1.60 ml, 3.20 mmol) was added slowly to a solution of 12 (1.24 g, 5.32 mmol) in toluene at -10 °C. After stirring at the same temperature for 2 h, acetone (0.195 ml) was added and the mixture was stirred for additional 15 min. The reaction mixture was poured into 1M HCI (12.7 ml) cooled in ice bath and extracted with AcOEt. The combined extracts were washed successively with 0.1 M HCl, satd aq NaCl, satd aq NaHCOs, and satd aq NaCl, then dried over anhyd MgSO4. Filtration and concentration *in vacuo* gave a residue which was purified with column chromatography (SiOa, hexane-AcOEt 19:l) to give **9a** as a colorless oil (0.76 g, 88%). The IR and lH-NMR spectra of this sample *were* superimposable on those of **9a** obtained in a).

(S)-2-(t-Butyldimethylsilyloxy)propanal (9b). Treatments of 8b in the same manner as described for the preparation of **9a** from 8a, gave 9b as a colorless oil (0.707g, 76%) after purification with bulb-to-bulb distillation (bp 90 "C, 20 mmHg). IR (neat): 2952,2948,2852,1740,1255, 1135. 838. 778 cm⁻¹. ¹H-NMR (CDCls): 0.10 (6H, s, Me2Si), 0.92 (9H, s, MesC), 1.28 (3H, d, J=6.8 Hz, MeCH), 4.09 (1H, dq, J=1.3, 6.8 Hz, MeCH), 9.61 (1H, d, J=1.3 Hz, CHO). MS m/e: 159 (M-CHO)+, 131 (M-Bu)+, 103,73.

W-2-Tetrahydropyranyloxypropanal(9c). Treatments of 8c (0.445 g, **2.20** mmol) in the same manner as described for the preparation of **9a** from 8a, afforded 9c as a colorless oil (0.2838, 81%) after sequential purification with column chromatography (SiOz, hexane-AcOEt 19:1-8:l) and bulb-to-bulb distillation (110 °C/20 mmHg). IR (neat): 2950, 2870, 1738, 1378, 1125, 1080, 1038 cm^{-1} . ¹H-NMR (CDCl3): 1.32 (3H, d, J=7.1 Hz, Me), 1.5~1.9 (6H, m, (CH2)3), 3.4~4.0 (2H, m, b OCH2), 4.27 (1H, dq, J=1.2, 7.1 Hz, MeCH), 4.6~4.8 (1H, m, OCHO), 9.65, 9.66 (1H, two d, J=1.2 Hz, CHO). MS m/e: 84 $(THP)^{+}$, 55, 43.

CW4-(2.Methoxyethoxymethosy)propanal(9d). Treatmenta of 8d (l.O4g, 5.05 mmoI) in a similar_ manner to that described for the preparation of 9a from **8a,** afforded **9d** as a colorless oil (0.643 g, 78%) after sequential purification with column chromatography (SiOz, hexane-AcOEt 7:3-3:4) and distillation (bp 100 "C, 1 mmHg). IR (neat): 2950,2900,2830,1738,1450,1110,1038, 847 cm⁻¹. ¹H-NMR (CDCls): 1.32 (3H, d, J=7.0 Hz, <u>MeCH)</u>, 3.38 (3H, s, MeO), 3.48~3.59, 3.71-3.78 (4H, two m, OCzH40), 4.08 (lH, dq, J=1.5, 7.0 Hz, GHMe), 4.83 (2H, 8, OCHaO), 9.65 (1H, d, J=1.5 Hz, CHO). MS m/e: 133 (M-CHO)⁺, 119, 89.

(S)-2-(1-Ethoxyethoxy)propanal (9e). Reduction of 8e was performed by the same procedure as described for the preparation of θ a from θ a, affording θ e (R=EE) as a colorless oil (27.6g, 82 %) after distillation (bp 103 °C, 20 mmHg). ¹H-NMR (CDCls): 1.1~1.5 (9H, m, Mex3), 3.4~4.4 (3H, m, other protons), 4.7-5.0 (lH, m, OCHO), 9.61 (lH, t, Je2.4 Hz, CHO).

(S)-N_rN-Tetramethylenelactamide (11). Pyrrolidine (7.80 ml, 93.4 mmol) was added to 7 (10.2) g, 86.0 mmol) at 0° C and stirred at rt for 3 days. After removal of excess pyrrolidine and resulting ethanol *in vacua, the oily* residue was purified with distillation (108 "C, 1 mmHg) to give 11 (11.8g, 95%) as a colorless oil. $[\alpha]_D^{\text{20}}$ -49.2° (c 4.78, CHCl3). IR (neat): 3450, 3000, 2900, 1637, 1440, 1387, 1345, 1133, 1040 cm⁻¹. ¹H-NMR (CDCls): 1.33 (3H, d, J=6.6 Hz, Me), 1.49 (4H, m, NCH₂CH₂x₂), 3.45 (4H, m, NCH_{2x}₂), 3.73 (1H, d, J=7.3 Hz, OH), 4.29 (1H, dq, J=6.6, 7.3 Hz, CHCO). MS m/e: 143 (M)⁺, 128 (M-Me)⁺, 98 (M-MeCHOH)⁺. Found: C, 57.00; H, 9.22; N, 9.53%. Calcd for CrHieNOz: C, 56.93; H, 9.21; N, 9.49%.

(S)-N,N-Tetramethlene-2-benzyloxypropion amide (12). a) Preparation of 12 with benzyl chloride and sodium hydride. A solution of **11 (9.98 g, 67.7** mmol) in THF (15 ml) was added slowly to a suspension of sodium hydride $(2.00 \text{ g}, 83.3 \text{ mmol})$ in THF (50 ml) and DMF (25 ml) at 0 °C with vigorous stirring. After stirring for 4.5h at the same temperature, benzyl chloride (8.80 ml, 76.5 mmol) was added to the reaction mixture. Stirring was further continued overnight at 0 \degree C. The resulting mixture was diluted with H₂O (50 ml) and extracted with AcOEt. The combined extracts were washed with satd aq NaCl and dried over anhyd MgS04. After filtration and concentration *in vacua, the* residue was purified with crystallization from hexane-Et20 to give **12** as colorless crystals (14.1 g, 87%). Recrystallization from dibutyl ether gave an analytical sample of 12 as colorless crystals, mp $42-42.5$ °C and $[\alpha]_D^{\infty}$ -66.9° (c 1.72, CHCl3). IR (KBr): 3050, 3000, 2890, 1430, 1350, 1120, 730 cm⁻¹. ¹H-NMR (CDCls): 1.42 (3H, d, J=6.8 Hz, Me), 1.85 (4H, m, NCH₂CH₂x₂), 3.50 (4H, m, NCH₂x₂), 4.20 (1H, q, J=6.8 Hz, CHCO), 4.11, 4.62 (2H, two d, J=each 11.7 Hz, PhCH₂), 7.32 (5H, s, C6H₅). MS m/e: 234(M+1)⁺, 127, 98. Found: C, 72.06; H, 8.32; N, 5.90 %. Calcd for C14H19NO2: C, 72.07; H, 8.21; N, 6.00 %. b) Preparation of 12 under phase transfer conditions. Finely powdered sodium hydroxide $(0.737 \text{ g}, 17.5 \text{ mmol})$ and $11 (0.716 \text{ g}, 5.01 \text{ mmol})$ were added to a solution of benzyl chloride (0.760 m) g, 6.00 mmol) and tricapryhnethylammonium chloride (0.101 g, 0.250 mmol) in toluene (5 ml) at 0°C and the mixture was stirred vigorously at rt for 6h. The mixture was diluted with toluene and filtered. The filtrate was washed successively with 1M HCl (5 ml) and satd aq NaHCOs, then dried over anhyd MgS04. ARer filtration and concentration *in vacua,* the residue was purified with column chromatography (SiOz, hexane-AcOEt 2:3) to give 12 as col-

Di-p-anisylmethylamine (DAM-NH₂).¹⁹⁾ A mixture of 4,4'-dimethoxybenzophenone (9.29g, 37.0) mmol) and ammonium formate $(14.1g, 220 \text{ mmol})$ was stirred at $180 \sim 190 \degree \text{C}$ for 4h. Formamide $(5.00g, 111mmol)$ and anhyd MgCl₂ $(0.247 g, 2.59 mmol)$ were added to the mixture. After stirring at 180-190 "C for 2h, the mixture was cooled and dissolved in CH2Clz. The dichloromethane solution was washed with H20 and concentrated *in vucuo to give* crude Nformyl-di-p-anisylmethylamine $(10.3 g)$ as a solid. A 1 M solution of HCl in MeOH (55ml) was added to the solid and the mixture was heated at reflux for 2h, then concentrated *in vacua.* The residue was dissolved in Hz0 (50 ml) and *the* aqueous solution was washed with toluene. The aqueous layer was made alkaline with 50% aq NaOH and extracted with toluene. The organic extracts were combined and dried over anhyd MgSO4. Filtration and concentration in uacuo gave DAM-NH2 as slightly yellow crystals (8.48 g, 94%). An analytical sample was obtained by recrystallization from ether, mp $59.5-60$ °C. IR (KBr): 3010, 2960, 2850, 1605, 1508, 1242, 1173, 1025, 560 cm⁻¹. ¹H-NMR(CDCls): 1.70 (2H, s, NH₂), 3.77 (6H, s, MeOx2), 5.12 (1H, s,

orless crystals (1.07 g, 92%). This sample showed the same 1H-NMR spectrum and optical ro-

tation as that of 12 obtained in a).

CHAr2), 6.78~7.31 (8H, m, aromatics). MS m/e: 244 (M+1)⁺, 243 (M⁺), 227 (M-NH2)⁺, 212 (M-MeO)⁺, 135. Found: C, 73.99; H, 7.14; N, 5.65 %. Calcd for C15H17NO2: C, 74.05; H, 7.04; N, 5.76%.

(3S,4S)-3-Acetyl-4-[(S)-1-benzyloxyethyl]-1-(di-p-anisylmethyl)-2-azetidinone (13a) and Its (3R,4R)-Isomer (14a) (Table 1, run 2). General procedure of the [2+2]-cycloaddition reaction of diketene with 10. Anhyd MgSO4 (5.00 g, 41.5 mmol) and DAM-NH2 (8.08 g, 33.2 mmol) were added to a solution of 9a (5.45 g, 33.2 mmol) in toluene (15 ml) at 0 °C. After stirring at the same temperature for 1h, the mixture was filtered and the collected materials were washed with toluene. The combined filtrates were concentrated in vacuo to give almost pure 10a. $\mathrm{^{1}H\text{-}}$ NMR (CDCls): 1.36 (3H, d, J=6.4 Hz, MeCH), 3.78, 3.79 (6H, two s, MeOx2), 4.17 (1H, m, MeCH), 4.51 (2H, s, PhCH₂), 5.34 (1H, s, Ar₂CH), 6.7~6.9, 7.1~7.3 (8H, two m, C6H4x2), 7.28 (5H, s, C6H5), 7.70 (1H, d, J=5.5 Hz, N=CH). This was immediately used for the next step without further purification. Imidazole $(2.26 \text{ g}, 33.2 \text{ mmol})$ was added to a solution of 10a in CH2Cl2 (33 ml) . The mixture was cooled to -35 °C and diketene $(7.80 \text{ ml}, 99.6 \text{ mmol})$ was added. After stirring at the same temperature for 21h, an additional amount of diketene (5.20 ml, 66.4 mmol) was added to the reaction mixture. Stirring was further continued at the same temperature for 2d. The mixture was diluted with ether and H2O. After stirring at rt, the aqueous layer was separated. The organic layer was washed successively with H2O, 1M HCl, satd ag NaCl, 2M NaOH (to remove dehydroacetic acid)²¹⁾ and satd aq NaCl, then dried over anhyd MgSO4. Filtration and concentration in vacuo gave a crude mixture of 13a and 14a as a slightly yellow oil. The ratio of 13a to 14a could be calculated as 7.3:1 based on the ${}^{1}H\text{-NMR}$ spectrum of the mixture. The methyl groups of acetyl moieties of 13a and 14a appear as two singlets at 2.27 and 2.33 ppm with an integration ratio of 7.3:1 (vide infra). The concentrated residue was purified with column chromatography (SiO2, hexane-ether-CH2Cl2 5:4:1~4:5:0), affording pure 13a as a colorless oil $(4.87 \text{ g}, 31\%, 2 \text{ steps})$ from the more polar fraction, a mixture of 13a and 14a (7.95 g, 51%, 2) steps, 13a:14a=16:1), and pure 14a as colorless crystals (1.50 g, 9.1%, 2 steps) from the less polar fraction. The combined yield of 13a and 14a based on 9a was 91%. The desired product (13a) showed the following physical and spectral data. $[\alpha]_D^{20}$ -7.3° (c 1.48, CHCls). IR (neat): 2950. 1760, 1720, 1614, 1589, 1514, 1460, 1306, 1250, 1190, 1099, 1036, 828, 740, 702, 598 cm⁻¹, ¹H-NMR (400 MHz, CDCls): 1.10 (3H, d, J=6.3 Hz, MeCH), 2.27 (3H, s, MeCO), 3.47 (1H, dq, J=6.3, 7.0 Hz, MeCH), 3.77, 3.79 (6H, two s, MeOx2), 3.87 (1H, d, J=2.5 Hz, C3-H), 4.15 (1H, dd, J=2.5, 7.0 Hz, C₄-H), 4.14, 4.45 (2H, two d, J=each 11.2 Hz, PhCH₂), 5.76 (1H, s, Ar₂CH), 6.78~6.85, 7.07~7.33 (13H, m, aromatic protons). MS m/e: 473 (M⁺), 382 (M-Bn)⁺. The ¹H-NMR spectrum of this sample measured in the presence of the chiral shift reagent [Eu(hfc)3], clearly showed the methyl group of acetyl moiety as two singlets at 2.57 and 2.75 ppm in an intensity ratio of 98:2. Since dl-13a²² exhibited two singlets of equal intensity at 2.57 and 2.75 ppm, the optical purity of 13a was estimated as 96 %ee. Accordingly, it appeared evident that each synthetic steps to the stage of 13a could proceed without substantial racemization. The undesired product (14a) recrystallized from EtOH show the following physical and spectral data. Mp 84~85 °C and $[\alpha]_D^{20} + 30.0^{\circ}$ (c 1.21, CHCl3). IR (KBr): 2960, 2890, 2850, 1735, 1702, 1605, 1582, 1508, 1460, 1408, 1345, 1282, 1240, 1205, 1180, 1158, 1105, 1020, 818, 736, 690, 568, 532, 508 cm⁻¹. ¹H-NMR (400 MHz, CDCl3): 1.02 (3H, d, J=6.3 Hz, MeCH), 2.33 (3H, s, MeCO), 3.34 (1H, dq, J=1.9, 6.3 Hz, MeCH), 3.75, 3.78 (6H, two s, MeOx2), 4.01 (1H, dd, J=1.9, 2.4 Hz, C4-H), 4.07, 4.48 (2H, two d, J=each 11.5 Hz, PhCH₂), 4.36 (1H, d, J=2.4 Hz, C₃-H), 5.72 (1H, s, CHAr₂), 6.72~6.85, 7.10~7.40 (13H, two m, aromatic protons). MS m/e: 473 (M⁺), 430 (M-MeCO)⁺, 382 (M-Bn)⁺. Found: C, 73.54; H, 6.64; N, 3.17 %. Calcd for C29H31NO5: C, 73.55; H, 6.60; N, 2.96 %.

(3S,4S)-3-Acetyl-4-[(S)-1-(t-butyldimethylsilyloxy)ethyll-1-di-p-anisylmethyl-2-azetidinone (13b) and Its $(3R,4R)$ -Isomer (14b) (Table 1, run 10). Treatments of 9b $(0.594 \text{ g}, 3.15 \text{ mmol})$ in a similar manner to that described for the preparation of $10a$ from 9a gave $10b$ (1.31 g) after

concentration of the toluene solution. 1 H-NMR (CDCls): 0.03, 0.07 (6H, two s, Me2Si), 0.89 (9H, s, MesC), 1.33 (3H, d, J=6.4 Hz, MeCH), 3.80 (6H, s, MeOx2), 4.46 (1H, dq, J=4.8, 6.4 Hz, OCH), 5.32 (lH, s, CHArz)), 6.86, 7.21(8H, two d, J=each 8.5 Hz, aromatic protons), 7.68 (lH, d, J=4.8 Hz, N=CH). A part of the imine $(10b)$ $(0.412 g, 0.997 mmol)$ was subjected to the $[2+2]$ -cycloaddition reaction in **CH2Cla** according to the general procedure, giving a mixture of 13b and 14b as a colorless oil (0.339 g, 68%, 2 steps) after purification with column chromatography (SiOz, hexane-AcOEt-CH2Cl2 8:1:1-7:1:0). The ratio of 13b to 14b could be calculated as 5.4:1 by measuring the ¹H-NMR spectrum of the mixture *(vide infra)*. These diastereomers (13b and 14b) could be separated with preparative TLC (SiOz; CH2C12-AcOEt 97:3, two developments). The desired product (13b) obtained as a colorless oil showed the following physical and spectral data. $[\alpha]_0^{20}$ -19.1° (c, 0.51, CHCl3). IR (neat): 2940, 2910, 2850, 1748, 1703, 1602, 1500, 1240, 1165, 1020, 820, 765 cm⁻¹. ¹H-NMR (CDCls): -0.04, -0.08 (6H, two s, Me2Si), 0.82 (9H, s, Me3C), 0.99 $(3H, d, J=6.2 \text{ Hz}, \text{MeCH})$, 2.27 $(3H, s, \text{MeCO})$, 3.79 $(6H, s, \text{MeOx2})$, 3.39~3.58 $(1H, m, C3-H)$, 4.04-4.16 (2H, m, other protons), 5.75 (lH, s, CHArs), 6.80-6.90,7.13-7.26 (8H, two m, aromatic protons). MS m/e: 469 (M-CO)⁺, 310, 227. The undesired product (14b) obtained as a colorless oil showed the following physical and spectral data. $[\alpha]_{D}^{20}+10.8^{\circ}$ (c 0.91, CHCl3). IR (neat): 2950, 2850, 1748, 1710, 1610, 1510, 1242, 1168, 1030, 830, 775 cm⁻¹. ¹H-NMR (CDCls): 0.01, 0.02 (6H, two s, Me2Si), 0.90 (9H, s, Me3C), 0.98 (3H, d, J=6.2 Hz, MeCH), 2.31 (3H, s, MeCO), 3.77, 3.79 (6H. two 8. MeOx2). 3.95-4.16 (3H. m, other protons), 5.59 (lH, s, CHAr2). 6.77-6.90, $7.14 - 7.26$ (8H, two m, aromatic protons). MS m/e: 497 (M⁺), 469 (M-CO)⁺, 310, 227. The stereochemistries of the reaction products were determined by converting the separated samples of 13b and 14b to 16f and 17f, respectively (vide infra).

(3S,4S)-3-Acetyl-1-di-p-anisylmethyl-4-[(S)-1-tetrahydropyranyloxyethyl]-2-azetidinone (13c) and Its $(3R,4R)$ -Isomer (14c) (Table 1, run 12). Treatment of 9c $(0.408 g, 2.58 mmol)$ in a similar manner to that described for the preparation of 10a from 9a gave 10c $(1.02 g)$ after concentration of the toluene solution. ¹H-NMR (CDCl₃): 1.33, 1.36 (3H, two d, J=6.6Hz, Me), 1.2 \sim 1.9 (6H, m, (CH2)3), 3.3~4.8 (4H, m, other protons), 3.77 (6H, s, MeOx2), 5.32 (1H, s, CHAr2), 6.83, 7.19 (8H, two d, J=each 8.8 Hz, aromatic protons), 7.66, 7.78 (1H, two d, J=5.5, 4.8 Hz, N=CH). A part of the imine (10c) (0.278 g, 0.726 mmol) was subjected to the $[2+2]$ -cycloaddition reaction in CH2Cl2 according to the general procedure, giving a mixture of 13 c and 14 c as a colorless oil (0.241 g, 71%, 2 steps) after column chromatography (SiOz, hexane-AcOEt-CH2Cl2 7:2:1-5:2:0). IR (neat): 3510, 2950, 2850, 1750, 1710, 1610, 1582, 1515, 1285, 820 cm⁻¹. ¹H-NMR (CDCl3): 0.97, 1.02, 1.10, 1.15 (3H, four d, J=6.4 Hz, Me), 1.4-1.8 (6H, m, (CH2)3), 2.27, 2.28, 2.31, 2.34 (3H four s, MeCO), 3.3-3.7 (2H, m, OCH2), 3.79 (6H, s, MeOx2), 3.8-4.6 (10H m, other protons), 5.65, 5.66, 5.73,5.84 (lH, a, CHAra), 6.8-6.9, 7.1-7.3 (8H, two m, aromatics). MS m/e: 476 (MY, 424, 382, 227. The stereochemistries of the reaction products were determined by converting the mixture of 13c and 14c to that of 16f and 17f (vide infra). The ratio of 13c to 14c could be calculated as 6.0:1 by the lH-NMR spectrum of the mixture of 16f and 17f.

(3S,4S)-3-Acetyl-1-di-p-anisylmethyl-4-[(S)-1-(2-methoxyethoxymethoxy)ethyl]-2-azetidinone

(13d) and Its $(3R,4R)$ -Isomer (14d) (Table 1, run 13). The same treatments of 9d $(0.594 \text{ g}, 3.15$ mmol) as described for the preparation of 10a from $9a$ gave 10d $(1.21g)$ after concentration of the toluene solution. ¹H-NMR (CDCl₃): 1.35 (3H, d, J=6.6 Hz, <u>Me</u>CH), 3.34 (3H, s, <u>Me</u>OCH2), 3.38~3.73 (4H, m, OC2H5O), 3.77 (6H, s, MeOArx2), 4.42 (1H, m, CHMe), 4.76 ((1H, s, CHAr2), 6.83, 7.18 (8H, two d, J=each 8.8 Hz, aromatic protons), 7.70 (1H, d, J=4.8 Hz, N=CH). A part of the imine (10d) (0.489 g, 1.26mol) was subjected to the $[2+2]$ -cycloaddition reaction in CH2Cl2 according to the general procedure, giving a mixture of 13d and 14d as a colorless oil (0.519 g, 87%, 2 steps) after purification with column chromatography (SiOe, hexane-AcOEt-CHzCl2 6:4:1~1:1:0). The ratio of 13d to 14d could be calculated as 6.0:1 by the ¹H-NMR spectrum of the mixture (vide infra). IR (neat): 2950, 2850, 1758, 1718, 1610, 1510, 1245, 1180, 1038 cm⁻¹. ¹H-

NMR (CDCls): 1.04, 1.08 (3H, two d, J=6.3, 6.4 Hz, MeCH, intensity ratio; 1:6.0), 2.28, 2.32 (3H, two s, MeCO, intensity ratio; 6.0:1), 3.36, 3.37 (3H, two s, MeO, intensity ratio; 6.0:1), 3.4~4.8 (9H, m, other protons), 3.78, 3.80 (6H, two s, MeOArx2), 5.70 (1H, s, CHAr2), 6.8~7.0, 7.1~7.4 (8H, two m, aromatic protons). MS m/e: $471 (M⁺)$, $443 (M⁻CO)⁺$, 428 , 227 . The stereochemistries of the reaction products were determined by converting the mixture of 13d and 14d to that of 16d and $17d$ (vide infra).

(3S,4S)-3-Acetyl-1-di-p-anigylmethyl-4-[(S)-1-(1-ethoxyethoxy)ethyll-2-azetidinone (13e) and Its $(3R,4R)$ -Isomer (14e) (Table 1, run 14). The same treatments of $9e(1.10g, 7.53$ mmol) as described for the preparation of 10a from 9a gave 10e after concentration of the toluene solution. ¹H-NMR (CDCls): 0.96~1.41 (9H, m, Mex3), 3.2~4.5 (3H, m, other protons), 3.77 (6H, s, MeOx2), 4.70 (1H, q, J=5.3 Hz, OCHO), 5.31 (1H, s, CHAr2), 6.7~7.0, 7.1~7.3 (8H, m, aromatic protons), 7.64, 7.72 (1H, two d, $J=5.4$ Hz, N=CH). The imine (10e) was directly subjected to the $[2+2]$ cycloaddition reaction in CH2Cl2 following the general procedure, giving a mixture of 13e and 14e as a colorless oil $(2.34 \text{ g}, 76\%, 2 \text{ steps})$ after purification with column chromatography (SiO2; CH2Cl2-acetone 98:2). ¹H-NMR (CDCl3): 0.9~1.3 (9H, m, Mex3), 2.27, 2.30, 2.32 (3H, three s, MeCO, intensity ratio; 7.4:1.0:1.5), 3.2~4.8 (6H, m, other protons), 3.78 (6H, s, MeOx2), 5.65, 5.73. 5.79 (1H, three s, intensity ratio; 1,0;2,0;1,5), 6.8~7.0, 7.1~7.3 (8H, two m, aromatic protons). The stereochemistries of the reaction products were determined by converting the mixture of 13e and 14e to that of 16f and 17f (vide infra). The ratio of 13e to 14e could be calculated as 3.0:1 by the ¹H-NMR spectrum of the mixture of 16f and 17f.

(3S,4S)-4-[(S)-1-Benzyloxyethyl]-1-di-p-anisylmethyl-3-(1,1-dimethoxyethyl)-2-azetidinone (16a) and Its (3R,4R)-Isomer (17a). a) Preparation from 13a and 14a. Camphorsulfonic acid (6.0 mg, 0.026 mmol) and trimethyl orthoformate (0.50 ml, 4.6 mmol) were added to a solution of 13a (43.9 mg, 0.093 mmol) in MeOH (1.0 ml) at rt. After stirring at rt for 1.5h, the mixture was diluted with H2O and satd aq NaHCOs, and extracted with ether. The combined ethereal extracts were washed with satd aq NaCl and dried over anhyd MgSO4. Filtration and concentration in vacuo gave an oily residue which was separated with column chromatography (SiO2, hexane-AcOEt-CH2Cl2 8:2:1) to give 16a as a colorless oil (42.8 mg, 89%), $[a]_D^{20} + 4.2^{\circ}$ (c 2.14, CHCls). IR (neat): 2950, 2850, 1755, 1618, 1517, 1245, 1038, 738, 700 cm⁻¹. ¹H-NMR (CDCls): 1.16 (3H, d, J=5.9 Hz, MeCH), 1.27 (3H, s, MeC(OMe)2), 3.17 (1H, m, C3-H), 3.22 (6H, s, (MeO)2C), 3.5~3.7 (2H, m, C4-H and CHOBn), 3.77, 3.78 (6H, two s, MeOArx2), 4.13, 4.48 (2H, two d, J=each 11.4 Hz, CH2Ph), 5.77 (1H, s, CHAr2), 6.74~6.87, 7.16~7.24 (13H, two m, aromatic protons). MS m/e: 460 (M-MeO-CO)⁺, 444 . Treatments of $14a$ (56.9 mg, 0.120 mmol) in the same manner as described for the preparation of 16a from 13a gave 17a as a colorless oil (50.0 mg, 80%), α le²⁰ +2.2° (c 1.16, CHCl₃). IR (neat): 2950, 2850, 1748, 1618, 1517, 1242, 735, 700 cm⁻¹. ¹H-NMR (CDCls): 1.11 (3H, d, J=6.5 Hz, MeCH), 1.13 (3H, s, MeC(OMe)2), 3.24, 3.26 (6H, two s, (MeO)2C), 3.5~3.7 (3H, m, C3-H, C4-H, and CHOBn), 3.75, 3.77 (6H, s, MeOArx2), 4.17, 4.55 (2H, two d, J=each 11.7 Hz, CH2Ph), 5.70 (1H, s, CHAr2), 6.69~6.85, 7.15~7.34 (13H, two m, aromatic protons). MS m/e: 460 (M-MeO-CO)⁺, 444, 268, 227.

b) Preparation from 16f and 17f. Tricaprylmethylammonium chloride (2.4 mg, 0.006 mmol), powdered NaOH (18 mg, 0.43 mmol), and benzyl chloride (16 µl, 0.14 mmol) were added to a solution of 16f (50.5 mg, 0.12 mmol) in toluene (0.3 ml) at 0 \degree C and the mixture was vigorously stirred at rt for 12h. Additional amounts of powdered NaOH (20mg, 0.48 mmol) and benzyl chloride (16 µl, 0.14 mmol) were added to the reaction mixture and stirring was continued for 2.5 h. The mixture was diluted with H₂O and ether. The ethereal layer was washed successively with 0.2 M HCl, satd aq NaHCOs, and satd aq NaCl, then dried over anhyd MgSO4. Filtration and concentration in vacuo gave an oily residue which was purified with column chromatography (SiO₂, hexane-AcOEt-CH₂Cl₂ 8:2:1) to give 16a as an oil (41.3 mg, 68%). The IR and ${}^{1}H$ -NMR spectra of this sample were superimposable on those of 16a obtained in a). The

same treatments of 17f (17.3 mg, 0.040 mmol) as described above, gave 17a as a colorless oil (9.1 mg, 44 %) after purification with column chromatography. The ¹H-NMR spectrum of this mg, 44 %) after purification with column chromatography. sample were identical with those of 17a obtained in a).

(3S,4S)-1-Di-p-anisylmethyl-3-(1,1-dimethoxyethyl)-4-[(S)-1-(2-methoxyethoxymethoxy)ethyll-2azetidinone (16d) and Its (3R,4R)-Isomer (17d). a) Preparation from 13d and 14d. The same treatments of the mixture of 18d and 14d (65.7 mg, 0.139 mmol) as described for the preparation of 16a from 13a afforded a mixture of 16d and 17d as an oil (62.3 mg, 86%) after purification with column chromatography. By comparing the ${}^{1}H$ -NMR spectrum of this mixture with those of the pure samples of 16d and 17d prepared in b), it appeared evident that the major component involved in the mixture was 16d.

b) Preparation from 16f and 17f. N,N-Diisopropylethylamine (22 μ l, 0.13 mmol) and 2methoxyethoxymethyl chloride (14.2 μ l, 0.124 mmol) were added to a solution of 16f (36.1 mg, 0.084 mmol) in CH₂Cl₂ (0.2 ml) at 0 °C. After stirring at rt for 12h, additional amounts of N_Ndiisopropylethylamine $(22\mu l, 0.126$ mmol) and 2-methoxyethoxymethyl chloride $(14.2 \mu l, 0.124$ mmol) were added to the mixture. After stirring was further continued for 12h, the reaction mixture was diluted with H20 and ether. The ethereal solution was washed successively with satd aq NaHCO3, 0.2M HCl, and satd aq NaCl, then dried over anhyd MgSO4. Filtration and concentration in vacuo gave a residue which was purified with column chromatography $(SiO₂,$ hexane-ether 1:6), giving 16d as a colorless oil (35.0 mg, 80%). lR (CHC13): 2950, 2850, 1740, 1610, 1510, 1242, 1030, 750 cm⁻¹. ¹H-NMR (CDCl₃): 1.15 (3H, d, J=5.9 Hz, <u>Me</u>CH), 1.26 (3H, s, $MeC(OMe)z$, 3.18 (1H, m, C₃-H), 3.22 (6H, s, (MeO)₂C), 3.35 (3H, s, MeOCH₂), 3.45~3.70 (6H, m,</u> other protons), 3.78 (6H, 8, MeOArx2), 4.44,4.58 (2H, two d, J=each 7.0 Hz, OCHzO), 5.73 (lH, s, CHArz), 6.77-6.89, 7.19-7.33 (8H, two m, aromatic protons). MS m/e: 489 (M-CO)+, 442, 428. Sodium hydride (1 mg, 0.04 mmol) and methoxyethoxymethyl chloride (1 ml, 0.009 mmol) were added to a solution of 17f (1.3 mg, 0.030 mmol) in THF (0.2 ml) at 0 °C. After stirring at the same temperature for 3 h and at rt overnight, the mixture was diluted with Hz0 and extracted with ether. The ethereal layer was separated and treated in the same manner as described above, gave 17d as a colorless oil (1.5 mg, 94%) after purification with colunm chromatography (SiOz, hexane-ether 1:6-1:7). lR (CHCl3): 2950,2850, 1740, 1608,1508,1200, 1032,720 cm-'. 'H-NMR (CDCl3): 1.13 (3H, d, J=6.2 Hz, MeCH), 1.25 (3H, s, MeC(OMe)2), 3.25, 3.27 (6H, two s, (MeO)2C), 3.33 (1H, m, C3-H), 3.36 (3H, s, MeOCH2), 3.45~3.65 (6H, m, other protons), 3.78 (6H, 8, MeOArx2), 4.47, 4.67 (2H, two d, J=each 7.2 Hz, OCH2O), 5.62 (1H, s, CHArz), 6.78~6.88, 7.18-7.31(8H, two m, aromatic protons). MS m/e: 489 (M-CO)+, 442,428,227.

(3S,4S)-1-Di-p-anisylmethyl-3-(1,1-dimethoxyethyl)-4-[(S)-1-hydroxyethyl]-2-azetidinone (16f) and Its $(3R,4R)$ -Isomer (17f). a) Preparation from 13b and 14b. Treatments of 13b (32.1 mg, 0.065 mmol) in the same manner as described in b) afforded 16f as a colorless oil (17.5 mg, 63%) after purification with column chromatography. The diastereomer (14b) (14.0 mg, 0.028 mmol) was similarly treated under acetalization conditions, giving 17f (1.8 mg, 15%) after purification with column chromatography. 'H-NMR spectra of these compounds were identical with those of 16f and 17f obtained in b), respectively.

b) Preparation from 13c and 14c. Camphorsulfonic acid $(9.4 \text{ mg}, 0.040 \text{ mmol})$ and trimethyl orthoformate (1.00 ml, 9.14 mmol) were added to a solution of the mixture of 13c and 14c (0.101 g, 0.216 mmol) in MeOH (3.0 ml) at rt. After stirring for 1.5 h, the mixture was diluted with Hz0 and satd aq NaHC03, then extracted with ether. The combined ethereal extracts were washed with satd aq NaCl and dried over anhyd MgS04. Filtration and concentration *in vacua* gave a crude mixture of 16f and 17f as an oil. Based on the ${}^{1}H\text{-NMR}$ spectrum of this sample, the ratio of 16f and 17f could be calculated as 6.O:l. The mixture was separated with column chromatography (SiO2, CH2Cl2-AcOEt 4:1), giving the major product $(16f)$ as a colorless oil (73.8) mg, 79%) from the less polar fraction, $\lceil \alpha \rceil^{20} - 3.8^\circ$ (c 1.41, CHCl3). IR (CHCl3): 3000, 2850, 1742,

1618, 1518, 1250, 1040 cm⁻¹. ¹H-NMR (CDCls): 1.13 (3H, d, J=6.2 Hz, <u>Me</u>CHOH), 1.26 (3H, s. MeC(OMe)z), 1.4~1.6 (1H, br, OH), 3.16 (1H, d, J=2.3 Hz, C3-H), 3.21, 3.24 (6H, two s, (MeO)zC), 3.56 (1H, dd, J=2.3, 6.5 Hz, C4-H), 3.78, 3.80 (6H, two s, MeOArx2), 6.8~7.0, 7.2~7.4 (8H, two m, aromatic protons). MS m/e: 429 (M⁺), 268, 227. The minor product (17f) was also obtained as a colorless oil (10.0 mg, 11%) from the more polar fraction, α b^{20} +8.1° (c 0.76, CHCls). IR (CHCls): 3000, 2850, 1742, 1618, 1518, 1250, 1038 cm⁻¹. ¹H-NMR (CDCls) 1.07 (3H, d. J=6.4Hz, MeCH), 1.36 (3H, s, MeC), 1.5~1.6 (1H, br, OH), 3.27 (6H, s, C(OMe)z), 3.42 (1H, d, J=2.2 Hz, C3-H), 3.6~3.8 (2H, m, MeCHOH and C4-H), 3.79, 3.80 (6H, two s, MeOArx2), 5.84 (1H, s, CHArz), 6.8-7.0,7.2-7.4 (8H, two m, aromatic protons). MS m/e: 429 (M+), 268, 227.

c) Preparation from 13e and 14e. The same treatments of the mixture of 13e and 14e (0.100 g) 0.220 mmol) as described in b), gave a mixture of 16f end 17f ae an oil (88.0 mg, 97%) after purification with column chromatography. The ratio of 16f to 17f could be calculated as $3.0:1$ by the 'H-NMR spectrum of the mixture.

 $(3S,4S)-4-[S)-1-Benzyboxyethy]-1-di-p-anisylmethy]-3-[(R)-1-hydroxyethy]-2-azetidinone and Its$ $3-[S]-1-Hydroxyethyl-isomer (18 and 19).$ a) Preparation of 18 and 19 by the reduction with potassium tri-sec-butylborohydride in the presence of potassium iodide.²⁴⁾ A 1.0 M solution of potassium tri-sec-butylborohydride in THF (0.84ml, 0.84 mmol) was added to a suspension of 13a (0.197g, 0.416 mmol) and potassium iodide (69.0 mg, 0.416 mmol) in THF (4.2 ml) at 0 °C. After stirring at the same temperature for lh, the reaction mixture was diluted with 1M HCl (0.84 ml) and extracted with AcOEt. The combined extracts were washed with satd aq NaCl, dried over anhyd MgSO4, filtered, then concentrated in vacuo. The concentration residue was purified with column chromatography (SiO2, hexane-ether 1:9~0:1) to give a mixture of 18 and 19 as a colorless caramel (0.182 g, 92 %) with recovery of the starting material (13a) (7.6 mg, 4%). The ratio of 18 to 19 could be calculated as 12:1 based on the ${}^{1}H\text{-NMR}$ spectrum of the mixture measured in CsDs. The methyl groups of benzyloxyethyl moieties appeared as two doublets at 0.86 and 0.94 ppm with an integration ratio of 12:1. The physical and spectral data of 18 and 19 separated with column chromatography were described in b)

b) Preparation of 18 and 19 by the reduction with potassium triethylborohydride.²⁵⁾ A 1.0 M solution of potassium triethylborohydride in THF $(0.75 \text{ ml}, 0.75 \text{ mmol})$ was added to a solution of 13a (0.321g, 0.678 mmol) in THF (6.8 ml) at -78 °C. After stirring at the same temperature for lh, the reaction mixture was diluted witb 1M HCl (0.7 ml), stirred at rt for lh, and extracted with AcCEt. The combined extracts were washed with satd aq NaCl, dried over anhyd MgsO4, filtered, then concentrated *in vacuo*. The concentration residue was purified with column chromatography (SiO₂, CH₂Cl₂-acetone 1:0~9:1) to give a mixture of 18 and 19 as a colorless caramel (0.316 g, 98 %). The ratio of 18 to 19 could be calculated as 12:l in a similar manner to that described in a). Separation of the mixture of 18 and 19 with medium pressure column chromatography (SiO2, lobar column, Merck art. 10401, CH2Cl2-AcOEt 9:1) afforded pure 18 and 19 both as colorless crystals. Recrystallization of 18 from hexane-AcCEt or isopropanol gave an analytical sample as colorless crystals, mp $102-102.5$ °C and $[\alpha]_p^{20}$ -11.5° (c 1.03, CHCl₃). IR (CHCl₃): 3300~3500 br, 2850~3000 br, 1740, 1618, 1518, 1250, 1038 cm⁻¹, ¹H-NMR (CDCl3): 1.13 (3H, d, J=5.9 Hz, MeCHOBn), 1.25 (3H, d, J=6.4 Hz, MeCHOH), 2.0 (1H, br, OH), 2.86 (lH, dd, J=2.2,5.9 Hz, Cs-H), 3.49 (lH, m, CHOBn), 3.70 (lH, m, C4-H), 3.76,3.78 (6H two 8, MeOx2), 3.9~4.2 (1H, m, CHOH), 4.12, 4.47 (2H, two d, J=11.4 Hz, PhCH₂), 5.77 (1H, s, CHAr2), 6.7~7.4 (13H, m, aromatic protons). ¹H-NMR (C6Ds): 0.91 (3H, d, J=6.2 Hz, MeCHOBn), 1.16 (3H, d, J=6.3 Hz, MeCHOH), 1.90 (1H, br, OH), 2.59 (1H, dd, J=2.3, 6.3 Hz, C3-H), 3.26 (1H, m, $CL(OBn)$, 3.27, 3.29 (6H, two s, MeOx2), 3.69 (1H, dd, J=2.3, 7.5 Hz, C4-H), 3.90 (1H, m, MeCHOH), 3.91, 4.22 (2H, two d, J=each 11.7 Hz, PhCH₂), 5.97 (1H, s, CHAr2), 6.78, 7.36 (8H, two m, C6H4x2), 7.16 (5H, s, C6H5). MS m/e: 475 (M⁺), 447 (M-CO)⁺, 384 (M-Bn)⁺, 356. Found: C, 73.05; H, 7.22; N, 3.14 %. Calcd for C29H33NO5: C, 73.24; H, 6.99; N, 2.95%. The undesired product (19) was recrystallized from hexane-AcOEt to give an analytical sample as colorless

crystals, mp 127 °C and $[\alpha]_p^{28}$ +20.0° (c 0.15, CHCls). IR (CHCls): 3400~3600 br, 2850~3000 br, 1740, 1610, 1515, 1242, 1038 cm⁻¹. ¹H-NMR (CDCls): 1.14 (3H, d, J=5.9 Hz, MeCHOBn), 1.25 (3H, d, J=6.3 Hz, MeCHOH), 2.35 (1H, br, OH), 2.82 (1H, dd, J=2.0, 6.9 Hz, C3-H), 3.58 (1H, m, $CHOBn$, 3.64 (1H, dd, J=2.3, 7.0 Hz, C4-H), 3.78, 3.79 (6H, two s, MeOx2), 3.7~4.2 (1H, m, MeCHOH), 4.13, 4.48 (2H, two d, J=11.4 Hz, PhCH2), 5.77 (1H, s, CHAr2), 6.7~7.4 (13H, m, aromatic protons). ¹H-NMR(CsDs): 0.81 (3H, d, J=6.2 Hz, MeCHOBn), 1.10 (3H, d, J=6.3 Hz, MeCHOH), 1.30 (1H, br, OH), 2.55 (1H, dd, J=2.4, 6.6 Hz, C3-H), 3.22 (1H, m, CHOBn), 3.27, 3.29 (6H, two s, MeOx2), 3.43 (1H, dd, J=2.4, 7.9 Hz, C₄-H), 3.83 (1H, m, MeCHOH), 3.90, 4.21 (2H, two d, J=each 11.5 Hz, PhCH₂), 5.96 (1H, s, CHAr₂), 6.78, 7.36 (8H, m, C6H \star 2), 7.16 (5H, s, C6Hs). MS m/e: 475 (M⁺), 447 (M-CO)⁺, 384 (M-Bn)⁺, 340. Found: C, 72.32; H, 7.01; N,3.12 %. Calcd for $C29H33NO5 0.3H2O: C$, 72.42; H, 7.04; N, 2.91 %. The pure sample of 18 could be directly obtained in 76% yield by a single crystallization of the crude reduction product from isopropanol. In another experiment, a crude diastereomeric mixture of 13a and 14a (6.90 g) produced under the best conditions of the [2+2]-cycloaddition reaction (Table 1, run 2), was reduced with potassium triethylborohydride (19.0 ml, 19.0 mmol) without separation. When the mixture of the crude reduction products (7.00 g) was immediately recrystallized from isopropanol (7.0 ml) , an almost pure sample of 18 (3.23 g) could be obtained in 58% overall yield based on 9a [1.9Og, 11.6 mmol (calculated by taking into account the amount of 12 (3.08 g, 13.2 mmol) used for preparation of 9a and the chemical yield of 9a from 12)l.

c) Preparation of 18 by the epimerization of 19.²⁶⁾ Diethyl azodicarboxylate $(6.6 \mu l, 0.042 \text{ mmol})$ was added to a solution of 19 (5.7 mg, 0.012 mmol), triphenylphosphine (11 mg, 0.042 mmol), and formic acid (1.9 μ l, 0.050 mmol) in THF (0.2 ml) at 0 °C. After stirring at rt for 2h, the mixture was diluted with phosphate buffer (pH 5, 0.5 ml) and extracted with AcOEt. combined extracts were dried over anhyd MgSO₄ and concentrated *in vacuo*. The concentration residue was purified with preparative TLC (SiO2; CH2Cl2-AcOEt 9:1) to give $(3S,4S)$ -4-[(S) -1benzyloxyethyl]-1-(di-p-anisylmethyl)-3- $[(R)-1$ -formyloxyethyl]-2-azetidinone (5.5 mg, 91%). ¹H-NMR (CDCls): 1.16 (3H, d, J=6.0Hz, MeCHOBn), 1.36 (3H, d, J=6.4 Hz, MeCHOCO), 2.97 (1H, dd, J=1.4, 7.1 Hz, C4-H), 3.59 (2H, m, CHOBn and C4-H), 3.78, 3.79 (6H, two s, MeOx2), 4.10, 4.47 $(2H, two d, J=each 11.3 Hz, PhCH₂), 5.27 (1H, m, CHOCO), 5.76 (1H, s, CHAr₂), 6.72-7.31 (13H,$ m, aromatic protons), 7.90 (1H, s, HCOO). Anhyd K2CO3 (2 mg) was added to a solution of the formate in methanol (2 ml) and the mixture was stirred at rt for 10 min. Filtration and concentration *in vacua* followed by purification with column chromatography (SiOz, ether), gave 18 (5.1 mg, 90% from 19). The ¹H-NMR spectrum of this sample were identical with those of 18 obtained in b).

~3S,gS)-P[(S)-l-Benzyloxyethyll-3-[(R)-l-hydroxyethyl]-2-azetidinone (20). A solution of CAN $(3.26 \text{ g}, 5.95 \text{ mmol})$ in H₂O-MeCN (1.9) (19.8 ml) was added to a solution of 18a $(0.943 \text{ g}, 1.98 \text{ m}$ mmol) in H2O-MeCN (1:9) (14.7 ml) at -10 °C. The mixture was stirred vigorously at the same temperature for 3h, then diluted with 2M NaOH (11.8 ml). After stirring at rt for 30 min, the mixture was filtered. The filtrate was neutralized to pH 8 with 1M HCl and extracted with CHC13. The combined organic extracts were dried over anhyd MgS04, filtered, and concentrated *in vacua. The* concentration residue was purified with column chromatography (SiOz, CH₂Cl₂-acetone 4:1 \sim 2:1) to give 20 as colorless crystals (0.461 g, 93%). Recrystallization from AcOEt gave an analytical sample of 20 as colorless crystals, mp 129~130 °C and $\left[\alpha\right]_{\alpha=0}^{\infty}$ +61.5° (c 1.45, CHCls). IR (KBr): 3380, 3000, 2900, 1735, 1475, 1452, 1380, 1361, 1335, 1240, 1179, 1136, 1104, 1093, 1069, 1001, 770, 665, 600, 560 cm⁻¹. ¹H-NMR (CDCl3): 1.25 (3H, d, J=5.9 Hz, MeCHOBn), 1.31 (3H, d J=6.4 Hz, MeCHOH), 2.86 (1H, m, C₃-H), 3.58 (2H, m, CHOBn and C4-H), 4.16 (1H, m, CHOH), 4.42, 4.70 (2H, two d, J=11.6 Hz, PhCH₂), 5.98 (1H, bs, NH), 7.25 (5H, s, C6H5). MS m/e: 249 (M⁺), 232 (M-OH)⁺, 204 (M-MeCHOH)⁺. Found: C, 67.45: H, 7.82; N, 5.51 %. Calcd for C14HlsNOz: C, 67.45; H, 7.68; N, 5.62 %.

(3S,4S)-4-[(S)-1-Benzyloxyethyl]-3-[(R)-(1-t-butyldimethylsilyloxy)ethyl]-2-azetidinone (21). A mixture of 20 (0.252g, 0.530 mmol), imidazole $(0.252$ g, 3.70 mmol), and t-butyldimethylchlorosilane (0.16 g, 1.47 mmol) in DMF (2.0 ml) was stirred at rt for 12h. The mixture was diluted with AcOEt and washed with H2O. The organic phase was separated, dried over anhyd MgSO₄, and concentrated *in vacuo*. The concentration residue was purified with column chromatography (SiO₂, CH₂C₁₂-acetone 1:0~19:1) to give 21 as a colorless oil (0.302 g, 97%), α ₁²⁵ **+32.5" (c** 2.37, CHCls). IR (neat): 3470,2950,2880,1760,1378,1259, 1140,1100,837,780,735,699 cm^{-1} . ¹H-NMR (CDCls): 0.07 (6H, s, Me2Si), 0.87 (9H, s, Me3C), 1.22 (3H, d, J=6.2 Hz, MeCHOSi), 1.25 (3H, d, J=5.9Hz, MeCHOBn), 2.74 (1H, m, C3-H), 3.52 (2H, m, CHOBn, C4-H), 4.17 (1H, m, CHOSi), 4.40, 4.68 (2H, two d, J=each 11.7 Hz, PhCH2), 5.87 (1H, bs, NH), 7.32 (5H, s, C6Hs). MS m/e: 363 (M⁺), 348 (M-Me)⁺, 306 (M-Bu)⁺.

 $(3S,4S)$ -3-[(R) -1-(t -Butyldimethylsilyloxy)ethyl]-4-[(S) -1-hydroxyethyl]-2-azetidinone (22). A mixture of 21 (45.0 mg, 0.124 mmol) and 10% Pd-C (4 mg) in AcCEt (3.0 ml) was stirred at rt for Id under a hydrogen atmosphere. The catalyst was filtered off and the filtrate was concentrated *in vacuo* to give 22 as a colorless caramel (33.8 mg, quantitative yield), $[\alpha]_2^{32}$ -11.5° (c 1.48, CHCl₃). IR (neat): 3300, 2950, 2870, 1742, 1463, 1375, 1255, 1140, 1098, 1054, 960, 834, 810, 777cm⁻¹. ¹H-NMR (CDCl3): 0.08 (6H, s, Me2Si), 0.88 (9H, s, Me3C), 1.23 (3H, d, J=6.2 Hz, <u>Me</u>CHOH), 1.25 $(3H, d, J=6.2 Hz, \text{MeCHOSi})$, 2.85 (1H, m, C3-H), 3.55 (1H, dd, J=2.2, 6.8 Hz, C4-H), 3.72 (1H, m, $CHOH$), 4.20 (1H, quint, J=6.2 Hz, CHOSi), 6.11 (1H, bs, NH). MS m/e: 258 (M-Me)⁺, 216 (M- Bu ⁺.

 $(3S,4S)$ -4-Acetyl-3- $[(R)-1-(t-butyldimethylsilyloxy)ethyl]-2-azetidinone (23)$. a) Preparation by the oxidation of 22 with chromium(VI) trioxide. Chromium(V1) trioxide (2.08 g, 2.08 mmol) was added slowly to pyridine (21 ml) . A solution of 22 $(0.570 \text{ g}, 2.08 \text{ mmol})$ in pyridine (5.0 ml) was added to the pyridine solution containing chromium(VI) trioxide at 0° C. The mixture was stirred at 30 °C for 1h and at 40 °C for 1.5 h, then diluted with 1M HCl and AcOEt. The organic layer was separated, washed with sat aq NaCl, dried over anhyd MgSO4, then concentrated *in uacuo.* The concentration residue was purified with column chromatography (SiOa, hexane-AcCEt 1:l) to give 23 as colorless crystals (0.523 g, 93%). Recrystallization from hexane gave an analytical sample of 23 as colorless crystals, mp $71-74$ °C and $[\alpha]_D^{26}$ -14.3° (c 0.57, CHCl3). IR (KBr): 3250, 2980, 2950, 2880, 1754, 1730, 1708, 1364, 1256, 1141, 1073, 1042, 961, 838, 811, 780 cm⁻¹. 1 H-NMR (CDCl3): 0.11 (6H, s, Me2Si), 0.90 (9H, s, Me3C), 1.31 (3H, d, J=6.4 Hz, MeCHOSi), 2.25 (3H, s, MeCO), 3.08 (lH, m, C3-H), 4.28 (2H, m, C4-H, CHOSi), 6.05 (lH, bs, NH). MS m/e: 256 (M-Me)⁺, 214 (M-Bu)⁺. Found: C,57,72; H, 9.46; N, 5.15%. Calcd for C13H25NO3Si, C, 57.53; H, 9.28; N, 5.16%.

b) Preparation by the oxidation of 22 with a combination of N-chlorosuccinimide, dimethyl sulfide, and triethylamine. Freshly recrystallized N-chlorosuccinimide (0.125 g, 0.936 mmol) was dissolved in toluene (5.0 ml) at 40 °C and the solution was cooled to rt. Dimethyl sulfide (58.1 mg, 0.936 mmol) was added to the toluene solution at rt and the formed suspension was cooled to 0° C. After stirring at the same temperature for 20 min, the suspension was further cooled to -25 °C. A solution of 22 (64.0 mg, 0.234 mmol) in toluene (0.9 ml) was added to the suspension. After stirring for 2.7h, triethylamine (0.136 ml, 0.976 mmol) was added to the reaction mixture. The mixture was gradually warmed up to rt, stirred for 15 min, then diluted with 1M HCl (1.0 ml). The organic layer was separated, dried over anhyd MgSC4, then concentrated *in uacuo. The* concentration residue was purified with column chromatography to give 23 as colorless crystals $(60.0 \text{ mg}, 95\%)$. This sample showed the same ¹H-NMR spectrum as that of 23 obtained in a).

c) Preparation by the oxidation of 26 with sodium peroxydisulfate. A solution of sodium peroxydisulfate in H20 (90%, 0.815 g, 3.08 mmol) and a solution of disodium hydrogenphosphate $(0.437 \text{ g}, 3.08 \text{ mmol})$ in H₂O (7.7 ml) were added slowly to a stirred solution of 26 $(0.153 \text{ g}, 0.308 \text{ mol})$ mmol) in acetone-H2O (9:1, 16.5 ml) under reflux. The mixture was heated at reflux with stirring for 40 min. After cooling, the mixture was concentrated in vacuo to remove acetone and the residual aqueous solution was extracted with AcOEt. The combined extracts were dried over anhyd MgSO4, filtered, then concentrated in vacuo. The concentration residue was purified with column chromatography (SiO2, CH2Cl2-AcOEt 1:0~9:1) to give 23 as a colorless solid $(74.4 \text{ mg}, 89 \%)$. This sample showed the same ¹H-NMR spectrum as that of 23 obtained in a).

(3S.4S)-4-[(S)-1-Benzyloxyethyl]-3-[(R)-1-(t-butyldimethylsilyloxy)ethyl]-1-di-p-anisylmethyl-2-

azetidinone (24). A mixture of 18 (0.804 g, 1.69 mmol), 4-(dimethylamino)pyridine (0.310 g, 2.54 mmol), and t-butyldimethylchlorosilane $(0.306g, 2.03$ mmol) in CH2Cl2 $(0.8$ ml) was stirred at rt for 1d. The reaction mixture was diluted with CH2Cl2 and washed successively with 1M HCl, satd aq NaHCOs, and satd aq NaCl. The organic phase was dried over anhyd MgSO4 and concentrated in vacuo. The concentration residue was purified with column chromatography (SiO2; hexane-AcOEt 4:1) to give 24 as colorless crystals (0.946 g, 95%). An analytical sample of 24 was prepared as colorless crystals by recrystallization from isopropanol. Mp 84~85°C and $[\alpha]_D^{26}+19.8^{\circ}$ (c 1.38, CHCl3). IR (KBr); 3080, 3040, 2980, 2940, 2860, 1742, 1610, 1510, 1390, 1250, 1040, 825 cm⁻¹. ¹H-NMR (CDCl3): -0.05, 0.03 (6H, two s, Me2Si), 0.82 (9H, s, Me3C), 1.16 (3H, d, J=5.7 Hz, MeCH), 1.22 (3H, d, J=6.2 Hz, MeCH), 2.75 (1H, dd, J=2, 6.2 Hz, C3-H), 3.4~3.8 (2H, m, CHOBn, C4-H), 3.76, 3.78 (6H, two s, MeOx2), 4.1 (1H, m, CHOSi), 3.99, 4.42 (2H, each d, J=11 Hz, PhCH₂), 5.81 (1H, s, CHAr₂), 6.7 ~ 7.0, 7.1 ~ 7.3 (13H, two m, aromatic protons). MS m/e: 589 (M)⁺, 561 (M-CO)⁺, 532 (M-Bu)⁺. Found: C, 71.40; 8.27; N, 2.33%. Calcd for C35H47NO5Si: C, 71.27; H, 8.03; N, 2.37%.

(3S,4S)-3-[(R)-1-(t-Butyldimethylsilyloxy)ethyl]-1-di-p-anisylmethyl-4-[(S)-1-hydroxyethyl]-2-azetidinone (25). A 5M solution of HCl in MeOH (85 μ l) and 5% Pd-C (0.425 g) were successively added to a solution of 24 $(2.51 g, 4.25 mmol)$ in toluene $(8.5 ml)$, and the mixture was stirred at 40 °C for 12h under a hydrogen atmosphere. The catalyst was filtered off and the filtrate was concentrated in vacuo. The concentration residue was purified with column chromatography (SiO2, hexane-AcOEt 7:3) to give 25 as a colorless caramel (2.08 g, 98%), $[\alpha]_D^{25}$ -29.1° (c 3.32, CHCl3). IR (KBr): 3400, 2980, 2940, 2870, 1730, 1610, 1510, 1250, 1030, 825 cm⁻¹. ¹H-NMR (CDCl3): -0.06, 0.01 (3H, s, Me2Si), 0.79 (9H, s, Me3C), 1.13 (3H, d, J=5.9 Hz, MeCH), 1.23 (3H, d, J=6.2 Hz, <u>MeCH), 2.71 (1H, dd, J=2, 6.5 Hz, C3-H), 3.64 (2H, m, C4-H, CHOH), 3.78, 3.81 (6H, two</u> s, MeOx2), 3.9 (1H, br, OH), 4.03 (1H, m, CHOSi), 5.98 (1H, s, CHAr2), 6.8~7.0, 7.1~7.5 (8H, two m, aromatic protons). MS m/e: 442 (M-Bu)⁺. Found: C, 67.24; H, 8.53; N, 2.76. Calcd for C28H41NO5Si: C, 67.30; H, 8.27; N, 2.80%.

 $(3S,4S)$ -4-Acetyl-3- $(0R)$ -1- $(t$ -butyldimethylsilyloxy)ethyll-1-di-p-anisylmethyl-2-azetidinone (26). The same treatments of 25 (1.45 g, 2.89 mmol) as described for the prepartion of 23 from 22 to give 26 as colorless crystals $(1.41 g, 98%)$ after purification with column chromatography (SiO2; $CH2Cl2$ -AcOEt 1:0~9:1). An analytical sample of 26 was prepared as colorless crystals by recrystallization from hexane. Mp 85 °C and α α ²⁰+27.4° (c 1.32, CHCl3). IR (KBr); 2980, 2940, 2860, 1763, 1730, 1612, 1512, 1248, 1032, 825 cm⁻¹, ¹H-NMR (CDCl3) 0.09, 0.10 (6H, two s. Me2Si). 0.90 (9H, s, Me3C), 1.23 (3H, d, J=6.2 Hz, MeCH), 1.83 (3H, s, MeCO), 2.93 (1H, m, C3-H), 3.77, 3.78 (6H, s, MeOx2), 4.11 (1H, d, J=2.4 Hz, C4-H), 4.22 (1H, m, CHOSi), 5.81 (1H, s, CHAr2), 6.77~7.31 (8H, m, aromatic protons). MS m/e: 497 (M⁺), 440 (M-Bu)⁺. Found: C, 67.43; H, 8.01; N, 2.79. Calcd for C28H39NO5Si: C, 67.57; H, 7.90; N, 2.81%.

 $(3R,4R)-4$ -Acetoxy-3-[(R) -1- $(t$ -butyldimethylsilyloxy)ethyl]-2-azetidinone (6) . a) Oxidation with *m*-chloroperbenzoic acid. *m*-Chloroperbenzoic acid $(90\%, 0.221 \text{ g}, 1.15 \text{ mmol})$ was added to a solution of 23 (16.3 mg, 0.0602 mmol) in AcOEt (2.0 ml). After stirring at 35 °C for 2h, the mixture was diluted with AcOEt and washed successively with aq NaHSO3, satd aq NaHCO3, and

satd ag NaCl. The organic phase was dried over anhyd MgSO4, filtered, then concentrated in vacuo. The residue was purified with column chromatography (SiO2, hexane-ether 3:2) to give 6 as colorless crystals (16.1 mg, 93%). Recrystallization from hexane gave an analytical sample of 6 as colorless crystals, mp $108-109$ °C and $\lbrack \alpha \rbrack p^{25}+47.8^{\circ}$ (c 0.56, CHCls) $\lbrack \text{lit.} , \text{12b} \rbrack$ mp $101-103$ °C and $[\alpha]_{D^{26}}$ +47.9° (c 1.00, CHCl₃); lit.,^{12f} mp 104~106 °C and $[\alpha]_{D^{26}}$ +48.8° (c 0.41, CHCl₃); lit.,^{12g)} mp 107-108 °C and α α ² +50° (c 0.5, CHCl₃)]. IR (KBr): 2950, 1787, 1746, 1235, 1164, 1080, 1040, 839, 779 cm⁻¹. ¹H-NMR (CDCls): 0.07 (6H, s, Me2Si), 0.87 (9H, s, Me3C), 1.26 (3H, d, J=6.4 Hz, MeCH), 2.10 (3H, s, MeCO), 3.18 (1H, dd, J=1.3, 3.5 Hz, C3-H), 4.22 (1H, dq, J=3.5, 6.4 Hz, CHOSi), 5.84 (1H,d, J=1.3 Hz, C4-H), 6.51 (1H, bs, NH). MS m/e: 230 (M-Bu)⁺. Found: C, 54.45; H, 8.88; N, 4.80%. Calcd for C13H25NO4Si: C, 54.32; H, 8.77; N, 4.87%.

b) Oxidation with monoperphthalic acid. Hydrogen peroxide (35%, 2.5 ml, 25 mmol) was added to a solution of Na2COs (2.29 g, 21.6 mmol) in H2O (10.8 ml) at 0 °C. Phthalic anhydride (3.20 g, 21.6 mmol) was added to the aqueous solution at -5 °C. After stirring at the same temperature for 1.5 h, the mixture was diluted with a solution of sulfuric acid $(2.25 g, 21.6 mmol)$ in H₂O $(4.5$ ml) and AcOEt (11 ml). The organic layer was separated, washed with satd aq NH4OH, and satd aq NaCl, then dried over anhyd MgSO4. Filtration gave a solution of monoperphthalic The ketone 23 $(0.410 \text{ g}, 1.51 \text{ mmol})$ was added to the solution of monoperacid in AcOEt. phthalic acid in AcOEt at rt. After stirring at the same temperature for 8h, the mixture was diluted with 20% aq NaHSO3 to precipitate colorless crystals. The crystals were filtered off, and the filtrate was washed with 2M NaOH and dried over anhyd MgSO4. Filtration and concentration in vacuo gave an oil which was purified with column chromatography (SiO2, CH2Cl2-AcOEt 1:0~9:1) to give 6 as colorless crystals (0.334 g, 77 %). This sample showed the same 1 H-NMR spectrum as that of 6 obtained in a).

(3S,4S)-3-[(R)-1-(t-Butyldimethylsilyloxy)ethyl]-1-di-p-anisylmethyl-4-[(S)-1-(p-toluenesulfonyloxy)ethyl]-2-azetidinone (27a). p-Toluenesulfonyl chloride (0.535 g, 2.81 mmol) was added to a solution of 25 (0.933 g, 1.87 mmol) in pyridine (2.0 ml) at 0 °C. After stirring at the same temperature for 12h, the reaction mixture was diluted with AcOEt and H2O. The organic layer was washed with satd aq NaCl, then dried over anhyd MgSO4. Filtration and concentration in vacuo gave an oily residue which was purified with column chromatography (SiO2, hexane-AcOEt 73:27), giving 27a as a colorless caramel (1.00 g, 82%), $[\alpha]_0^{25}$ -22.2° (c 2.51, CHCl3). IR (CHCl3): 2950, 1750, 1615, 1514, 1368, 1179 cm⁻¹. ¹H-NMR CDCl3): -0.04, 0.04 (6H, two s, Me2Si), 0.83 (9H, s, MesC), 1.11 (3H, d, J=6.4 Hz, MeCHOSi), 1.15 (3H, d, J=6.4 Hz, MeCHOTs), 2.43 (3H, s, MePh), 2.85 (1H, dd, J=2, 5 Hz, C3-H), 3.73 (1H, m, C4-H), 3.80 (6H, two s, MeOx2), 4.08 (1H, m, MeCHOSi), 4.46 (1H, m, MeCHOTs), 5.67 (1H, s, CHAr2), 6.83, 7.14 (8H, two m, C6H4x2), 7.27, 7.66 (4H, two d. J=each 8.4Hz, C6H4SO2). MS m/e: 653 (M⁺), 596 (M-Bu)⁺.

 $(3S,4S)-3-[R)-1-(t-Butyldimethylsilyloxy)ethy]]-4-[S]-1-(t-boluenesulfonyloxy)ethy]]-2-azetidinone$ (27b). This was prepared from 22 (0.217 g, 0.795 mmol) in a similar manner to that described for the preparation of 27a from 25. The p-toluenesulfonate (27b) obtained as colorless crystals $(0.289 \text{ g}, 85\%)$ after purification with column chromatography, showed mp 125-126 °C (isopropyl ether) and [α]²⁰-19.4° (c 1.55, CHCl3). IR (KBr): 3250, 2950, 1754, 1714, 1360, 1258, 1202, 1190, 1175, 1080, 990, 913, 841, 811, 773, 555 cm⁻¹. ¹H-NMR (CDCl₃): 0.05, 0.06 (6H, two s, Me2Si), 0.86 (9H, s, Me3C), 1.17 (3H, d, J=6.3Hz, MeCHOSi), 1.32 (3H, d, J=6.6 Hz, MeCHOTs), 2.45 (3H, s, MeC6H4), 2.73 (1H, m, C3-H), 3.68 (1H, m, C4-H), 4.16 (1H, m, CHOSi), 4.58 (1H, m, CHOSO2), 5.78 (1H, bs, NH), 7.35, 7.79 (4H, two d, J=each 8.4 Hz, C6H4). MS m/e: 370 (M-Bu)⁺. Found: C, 56.19; H, 7.81; N, 3.20; S, 7.60%. Calcd for C20H33NO5SSi; C, 56.17; H, 7.78; N, 3.28; S, 7.50%.

(3S,4S)-3-[(R)-1-(t-Butyldimethylsilyloxy)ethyl]-1-di-p-anisylmethyl-4-[(RS)-1-iodoethyl]-2-azetidinone (28a). Sodium iodide (0.830 g, 5.55 mmol) was added to a solution of $27a(0.727 g, 1.11)$ mmol) in acetone (4.5 ml). After heating at reflux for 12h, the reaction mixture was diluted with AcOEt and H2O. The organic layer was separated, washed with satd aq NaCl, then dried over anhyd MgSO4. Filtration and concentration in vacuo gave an oily residue which was purified with column chromatography (SiO2; hexane-AcOEt 4:1), giving 28a as a colorless oil $(0.607 \text{ g}, 90\%)$. The ¹H-NMR spectrum revealed that 28a consisted of an epimeric mixture with respect to the 1-iodoethyl group in a ratio of 5:2. IR (neat): 2950, 1755, 1610, 1585, 1510, 1460, 1375, 1302, 1250, 1175, 1033, 830, 776 cm⁻¹. ¹H-NMR (CDCls) 0.01, 0.06, 0.07, 0.10 (6H, four s. Me2Si, intensity ratio 2:5:2:5), 0.87, 0.91 (9H, two s, Me3C, intensity ratio 2:5), 1.26, 1.31 (3H, two d, J=6.4, 6.2 Hz, MeCHOSi, intensity ratio 5:2), 1.75, 1.77 (3H, two d, J=6.8, 7.0 Hz, MeCHI, intensity ratio 2:5), 2.96 (1H, m, C₃-H), 3.78 (3H, s, MeO), 3.79, 3.80 (3H, two s, MeO, intensity ratio 2:5), 3.8~4.3 (3H, m, C4-H, CHI, CHOSi), 5.67, 5.83 (1H, two s, CHAr2, intensity ratio 5:2), 6.87, 7.28 (8H, two m, aromatic protons). MS m/e: 552 (M-Bu)⁺.

(3S,4S)-3-[(R)-1-(t-Butyldimethylsilyloxy)ethyl]-4-[(RS)-1-iodoethyl]-2-azetidinone (28b). Treatments of 27b (0.124 g, 0.290 mmol) in the same manner as described for the preparation of 28a from 27a gave 28b as colorless crystals (0.101 g, 91%) after purification with column chromatography, mp 145~146 °C. This sample (28b) was found to consist of an epimeric mixture with respect to the 1-iodoethyl group in a ratio of 6:5 by the ¹H-NMR spectrum. IR (KBr): 3170, 3100, 2950, 1766, 1722, 1250, 1181, 1138, 1101, 1050, 960, 832, 778 cm⁻¹, ¹H-NMR (CDCla); 0.08 (6H, s. Me2Si), 0.88 (9H, s, Me3C), 1.25, 1.34 (3H, two d, J=6.2, 6.4 Hz, MeCHOSi), 1.91 (3H, d, J=6.8 Hz, MeCHI), 2.72, 2.89 (1H, two m, C3-H, intensity ratio 6:5), 3.7~3.9 (1H, m, C4-H), 4.0~4.3 (2H, m, CHI, CHOSi), 5.91 (1H, bs, NH). MS m/e: 368 (M-Me)⁺, 326 (M-Bu)⁺. Found: C, 40.72; H, 6.90; N, 3.54%. Calcd for C13H26IO2Si: C, 40.73; H, 6.84; N, 3.65%.

(3S,4R)-3-[(R)-1-(t-Butyldimethylsilyloxy)ethyl]-1-di-p-anisylmethyl-4-vinyl-2-azetidinone (29a). 1,8-Diazabicyclo[5.4.0]undec-7-ene (0.202 ml, 1.35 mmol) was added to a solution of 28a (0.411 g, 0.675 mmol) in toluene (2.7 ml). After stirring at 100 °C for 12 h, the mixture was concentrated. The concentration residue was purified with column chromatography (SiO₂, hexane-AcOEt 4:1~7:3) to give 29a as a colorless oil (0.294 g, 91%), $[\alpha]_D^{26} + 52.5^{\circ}$ (c 1.18, CHCl3). IR (neat): 2950, 1755, 1613, 1588, 1513, 1462, 1304, 1250, 1178, 1035, 830, 779 cm⁻¹, ¹H-NMR (CDCla): 0.02. 0.05 (6H, two s, Me2Si), 0.83 (9H, s, Me3C), 1.17 (3H, d, J=6.2Hz, MeCH), 2.88 (1H, dd, J=2, 5 Hz, C3-H), 3.79, 3.80 (6H, two s, MeOx2), 4.09 (2H, m, C4-H and CHOSi), 5.03 (1H, bd, J=9.5 Hz, trans-CH=CH2), 5.12 (1H, bd, J=7.5 Hz, cis-CH=CH2), 5.64 (1H, m, CH=CH2), 5.80 (1H, s, CHAr2), 6.83, 7.17 (8H, two m, aromatic protons). MS m/e: $424(M-Bu)^+$.

(3S,4R)-3-[(R)-1-(t-Butyldimethylsilyloxy)ethyl]-4-vinyl-2-azetidinone (29b). a) Preparation from 29a. A solution of CAN (0.851 g, 1.55 mmol) in H2O (5.0 ml) was added to a solution of 29a (0.249 g, 0.518 mmol) in CH3CN (5.2 ml) at -10 °C. After vigorous stirring at the same temperature for 1h, the mixture was diluted with 2M NaOH (1.5 ml) and extracted with AcOEt. The organic extracts were conbined, washed with satd aq NaCl, dried over anhyd MgSO4, then concentrated in vacuo. The concentration residue was purified with column chromatography (SiO2, CH2Cl2-AcOEt 1:0~9:1) to give 29b as colorless crystals (98.0 mg, 74%). An analytical sample of 29b was obtained by recrystallization from pentane, mp $63-64.5$ °C and $\left[\alpha\right]_{2}^{28}$ -24.5° (c 1.05, CHCls). IR (KBr): 2950, 1760, 1720, 1253, 1140, 1104, 1060, 1030, 964, 835, 805, 777 cm⁻¹. ¹H-NMR (CDCl3): 0.08 (6H, s, Me2Si), 0.88 (9H, s, Me3C), 1.21 (3H, d, J=6.4 Hz, MeCH), 2.87 (1H, m, C3-H), 4.0~4.4 (2H, m, C4-H, CHOSi), 5.15 (1H, d, J=9.8 Hz, cis-CH=CH2), 5.29 (1H, d, J=15.7 Hz, trans-CH=CH2), 5.96 (1H, ddd, J=6.8, 9.8, 15.7 Hz, CH=CH2), 5.98 (1H, bs, NH). MS m/e: 178 (M-Bu)⁺. Found: C, 60.96; H, 9.78; N, 5.43%. Calcd for C13H25NO2Si: C, 61.13; H, 9.86; N, 5.48%.

b) Preparation from 28b. Treatments of 28b (64.0 mg, 0.167 mmol) in the same manner as described for the preparation of 29b from 29a gave 29b as colorless crystals (10.1 mg, 24%) after purification with column chromatography. This sample showed the ¹H-NMR spectrum identical with that of 29b obtained in a).

(3S,4S)-3-[(R)-1-(t-Butyldimethylsilyloxy)ethyl]-4-(2-hydroxyethyl)-2-azetidinone (30). a) Hydroboration with diborane. A 1.0 M solution of borane in THF (0.10 ml, 0.10 mmol) was added to a solution of 29b (20.1 mg, 0.079 mmol) in THF (0.32 ml) at 0 $^{\circ}$ C. After stirring at rt for 1h, the mixture was diluted with H₂O (0.1 ml) , 2M NaOH (0.1 ml) , and 35% H₂O₂ (0.12 ml) , and extracted with CH2Cl2. The extracts were combined and dried over anhyd MgSO4. Filtration and concentration in vacuo gave an oily residue which was separated with preparative TLC (SiO₂, hexane-AcOEt 1:4) to give a mixture of 22 and its $4-[R]-1$ -hydroxyethyl]-isomer as an oil from the less polar fraction (3.8 mg, 18%) and 30 as colorless crystals from the more polar fraction (4.7 mg, 22%). The mixture of 22 and its $4-[R]-1$ -hydroxyethyl]-isomer showed the following ¹H-NMR spectrum. ¹H-NMR (CDCl₃): 0.08 (6H, s, Me2Si), 0.88(9H, s, Me3C), 1.18~1.28 (6H, m, MeCHOH, MeCHOSi), 2.85, 3.03 (1H, two m, Cs-H, integration ratio was 2:3), 3.55, 3.61 (1H, two dd, J=2.2, 6.8 Hz and J=2.1, 4.7 Hz, C4-H), 3.65~4.40 (2H, m, CHOH, CHOSi), 5.93, 6.20 (1H, two bs, NH, integration ratio was 3:2). Based on this ¹H-NMR spectrum, the ratio of 22 to its 4- $[(R)-1-hydroxyethyl]$ -isomer could be calculated as 2:3. The more polar product (30) exhibited the same ¹H-NMR spectrum as described in b).

b) Hydroboration with 9-borabicyclo[3.3.1]nonane. A 0.5 M solution of 9-borabicyclo[3.3.1] nonane in THF $(1.16 \text{ ml}, 0.58 \text{ mmol})$ was added to a solution of 29b $(74.5 \text{ mg}, 0.292$ mmol) in ether (0.6 ml) at rt. After stirring at rt for 3h, the mixture was diluted with 2M NaOH (0.3 ml), 35% H2O2 (0.25 ml), and EtOH (0.1 ml) at 0 °C. Stirring was continued at 0 °C for 20 min, then at rt for additional 10 min. The mixture was diluted with CH2Cl2 and the organic layer was separated, washed with satd aq NaCl, and dried over anhyd MgSO4. Filtration and concentration in vacuo gave an oily residue which was purified with column chromatography (SiO₂, CH₂Cl₂-MeOH 30:1) to give 30 as colorless crystals (61.5 mg, 77 %). An analytical sample of 30 was obtained by recrystallization from hexane. Mp 85~87 °C and $[\alpha]_D^{25}$ -22.3° (c 1.01, CHCl3). IR (KBr): 2950, 1732, 1258, 1098, 1079, 1065, 1042, 1023, 986, 958, 837, 775 cm⁻¹. ¹H-NMR (CDCl3): 0.10 (6H, s, Me2Si), 0.89 (9H, s, Me3C), 1.28 (3H, d, J=6.2 Hz, MeCH), 1.88 (2H, m, CH_2CH_2OH), 2.34 (1H, t, J=5.7 Hz, OH), 2.91 (1H, m, C3-H) 3.73 (3H, m, CH₂OH, C4-H), 4.16 (1H, m, CHOSi), 6.17 (1H, bs, NH). MS m/e: 258 (M-Me)⁺, 216 (M-Bu)⁺. Found: C, 57.05; H, 10.12; N, 4.99%. Calcd for C13H27NO3Si: C, 57.10; H, 9.95; N, 5.12%.

(3S,4R)-3-[(R)-1-(t-Butyldimethylsilyloxy)ethyl]-4-carboxymethyl-2-azetidinone (4). Ruthenium trichloride hydrate (1 mg, 0.005 mmol) was added to a two layer solution of 30 (61.5 mg, 0.225 mmol) and sodium metaperiodate (0.145g, 0.676 mmol) in CCl4 (0.9 ml), MeCN (0.9 ml), and H₂O (1.35 ml). The mixture was stirred vigorously at rt for 1h and diluted with CH₂Cl₂. The organic layer was separated, washed with satd NaCl, and dried over anhyd MgSO4. Filtration and concentration in vacuo gave a residue which was purified with column chromatography (SiO2, AcOEt) to give 4 as colorless crystals (44.4 mg, 69%). An analytical sample of 4 was obtained by recrystallization from hexane-AcOEt. Mp 150~154 °C (decomp.) and $[\alpha]_D^{20} + 16.1^{\circ}$ (c 0.69, CHCl3). IR (KBr): 3320, 2950, 1767, 1722, 1255, 1140, 1099, 1062, 1038, 968, 837, 780, 724 cm⁻¹. IR (CHCl3): 3340, 2950, 1740, 1260, 1144 cm⁻¹. ¹H-NMR (CDCl3): 0.07 (6H, s, Me2Si), 0.88 (9H, s, MesC), 1.21 (3H, d, J=6.2 Hz, MeCH), 2.65 (2H, m, CH2COOH), 2.81 (1H, m, C3-H), 3.95 (1H, m, C4-H), 4.18 (1H, quint, J=6 Hz, CHOSi), 6.0~7.4 (1H, br, COOH), 7.11 (1H, bs, NH). This ¹H-NMR spectrum was superimposable on that of the authentic sample.^{4f)} MS m/e: 272 (M-Me)⁺, 230 (M-Bu)⁺. Found: C, 54.38; H, 8.69; N, 4.94%. Calcd for C13H25NO4Si: C, 54.32; H, 8.77; N, 4.87%.

References and Notes

- 1) Part of this work has been the subject of two preliminary communications. Y. Ito, T. Kawabata, and S. Terashima, Tetrahedron Lett., 27, 5751 (1986). Y. Ito and S. Terashima, *C.&m. I&t., 445* (1987).
- 2) G. AIbersSch6nberg, B. H.A.rieon, 0. D. Hensens, J. Hirahtield, K. Hoogsteen, E. A. Kaczka, R. E. Rhodes, J. S. Kahan, F. M. Kahan, R. W. RatcIiffe, E. Walton, L. J. RuswinkIe, R. B. Morin, and B. G. Christensen, *J. Am. Chem.* Sot., **UK46491** (1978). J. S. Kahan, F. M. Kahan, R. Goegehnan, 5. A. Currie, M. Jackson, E. 0. Stapley, T. W. Miller, A. K. Miller, D. Hendlin, S. Mochales, S. Hernandez, H. B. Woodruff, and J. Birnbaum, J. Antibiot., 32,1 (1979).
- 3) a) W. Dückheimer, J. Blumbach, R. Lattrell, and K. H. Scheunemann, *Angeu. Chem., Int. Ed. Engl., 24,180 (1985).* b) T. Eametani, K Fukumoto, and M. Ihara, Heterocycles, 17,463 (1982). c) M. Shibuya, *Yuki Gosei &gaku Kyokai Shi,* 41,62 (1983). d) T. Nagahara and T. Kametani, *Heterocycles*, 25, 729 (1987). e) Y. Sugimura and T. Hiraoka, *Yakugaku Zasshi*, 107,175 (1987).
- 4) a) D. G. MeIiUo, I. Shinkai, T. Liu, K. Ryan, and M. Sletzinger, *Tetrahedron L&t.,* 21, 2783 (1980). b) D. G. MeIiIIo, T. Liu, K Ryan, M. Sletzinger, and I. Shinkai, *ibid.,* 22,913 (1981). c) I. Shinkai, T. Liu, R. A. Reamer, and M. Sletzinger, ibid., 23, 4899 (1982). d) M. Shiozaki, N. Ishida, T. Hiraoka, and H. Yanagisawa, *ibid.*, $22,5205$ (1981). e) N. Ikota, O. Yoshino, and K. Koga, *Chem. Pharm. Bull.*, 30, 1929 (1982). f) T. Iimori and M. Shibasaki, *Tetrahedron L&t., 26,1523* (1985).
- 5) a) K. Hirai, Y. Iwano, and K Fujimoto, *Tetrahedron I&t., 23,4025* (1982). b) A. Yoshida, Y. Tajima, N. Takeda, and 5. Oida, *ibid., 25,2793 (1984). c)* T. Chiba, M. Nagatsuma, and T. Nakai, *Chem. L&t..* 1343 (1985). d) H. Fhri and C.-P. Mak, *J. Org. Chem., 50, 3438* (1985). e) H. Maruyama, M. Shiozaki, and T. Hiraoka, *Bull. Chem. Sot. Jpn., 58, 3264* (1985). 9 K Fujimoto, T. Iwano and K Hirai, *ibid., 59,1887 (1986).*
- 6) As the compounds eqiuvalent to 6, the 4-phenylsulfonyl-, 4-methanesulfonyl, 4-chloro-, and 4benzoyloxy-2-azetidinone derivatives have also been synthesized. K Hirai, Y. Iwano, and K Fujimoto, *Tetrahedron L&t., 23,4025 (1982).* M. Shiozaki, N. Iahida, H. Maruyama, and T. Hiraoka, *Tetrahedron, 39,2399 (1983).* M. Endo, Can. *J. Ch.em., 65,214O* (1987).
- 7) T. Shibata, K. Iino, T. Tanaka, T. Hashimoto, Y. Kameyama, and Y. Sugimura, *Tetrahedron L&t., 26,4739* (1985).
- 8) Y. Nagao, T. Kumagai, S. Tamai, T. Abe, Y. Kuramoto, T. Taga, S. Aoyagi, Y. Nagase, M. Ochiai, Y. Inoue, and E. Fujita, *J. Am. Chem. Soc.*, 108, 4673 (1986). L. M. Fuentes, I. Shinkai, and T. N. Salzmann, ibid., 108, 4675 (1986). R. Deziel and D. Favreau, *Tetruhe&on L.&t., 27,5687* (1986). Y. Ito and S. Terashima, *ibid, 28,6625 (1987).* F. Shirai and T. Nakai, *J. Org. Chem.,* 52,5492 (1987). T. Shibata and Y. Sugimura, *The 107th Annual Meeting of The Pharmaceutical Society of Japan, April, 1987, Kyoto, Abstract Paper, p. 479.* A. Martel, J.-P. Daris, C. Bachand, J. Corbeil, and M. Menard, *Can. J. Chem., 66, 1537 (1988).*
- 9) D. H. Shih, F. Baker, L. Cama, and B. G. Christensen, *Heterocycles,* 21,29 (1984).
- 10) The *27th Zntersci. Conf: Antimicrob. Ag. Chemotherap., New* York, 1987, Abs.752-758.
- 11) a) T. Tanaka, T. Hashimoto, K Iino, Y. Sugimura, and T. Miyadera, *Tetrahedron L&t., 23, 3158 (1981).* b) K Fujimoto, Y. Iwano, K Hirai, and S. Sugawara, *Chem. Pharm. Bull., 34, 999 (1986).*
- 12) a) P.-J. Reider and E. J. J. Grabowski, *Tetrahedron L&t., 23, 2293 (1982).* b) M. Shiozaki, N. Ishida, H. Maruyama, and T. Hiraoka, *Tetrahedron, 39, 2399 (1983). c)* H. Maruyama, M. Shiozaki, and T. Hiraoka, *Bull. Chem. Sot. Jpn., 68, 3264 (1985).* d) A. Yoshida, T. Hayashi, N. Takeda, S. Oida, and E. Ohki, Chem. Pharm. Bull., 29, 2899 (1981). e) T. Chiba and T. Nakai, *Chem. Z&t., 651(1985X* 9 *Zdem, Tetrahedron Z&t., 26,4647 (1985). g)* T. Ohashi, K. Suga, I. Sada, T. Miyama, and K. Watanabe, *Japan Kokai Tokkyo Koho JP 61-*

18791 (1986). h) W. J. Leanza, F. DiNinno, D. . Muthard, R. R. Wilkening, K. J. Wildonger, R. W. Ratcliffe, and B. G. Christensen, *Tetrahedron*, **39**, 2505 (1983).

- 13) M. Sato, H. Ogasawara, E. Yoshimizu, and T. Kato, *Chem. Pharm. Bull.*, **31**, 1902 (1983).
- 14) M. Sunagawa, K Goda, M. Enomoto, and A. Sasaki, *Japan Kokai Tokkyo Z&ho JP 59-70663 0.984) and 59-70664 (1984).*
- 15) Recently, the [2+2]-cycloaddition reaction of a ketene derivative with a chiral aliphatic imine was reported. a) A. K. Bose, U. R. Hegde, D. R. Wagle, S. S. Bari, and M. S. Manhas, *J. Chem. Soc. Chem. Commun.*, 161 (1986). b) C. Hubschwerlen and G. Schmid, Helu. *Chim. Acta, 66,2206* **(1983).**
- 16) T. Iversen and D. R. Bundle, *J. Chem. Sot. Chem. Commun., 1240* (1981).
- 17) It was reported that treatment of (S)-benzyl lactate with benzyl bromide in the presence of sodium hydride resulted in considerable racemization. K Takai and C. H. Heathcock, *J. Org. Chem., 50,3247* **(1985).**
- 18) The preparation method of (S) -2-benzyloxypropanal explored in this study was also found to be applicable to the syntheses of various types of 2-alkoxypropanals (Y. Ito, Y. Kobayashi, and S. Terashima, to be published.).
- DAM-NH2 was prepared according to the procedure of Leukart-Wallach reaction (see Experimental). M. L. Moore, Org. React., 5, 301 (1949).
- 20) Recently, we have also found that the $[2+2]$ -cycloaddition reaction of diketene with the chiral imine derived from DAM-NH₂ and (S)-methyl 2-methyl-3-hydroxypropionate proceeds in a highly stereoselective manner. The produced optically active 3,4-trans-3-acetyl-6-lactam could be elaborated to $(3S,4S)-3-(R)-1-(t-butyldimethylsiyloxy)ethyl-4-(R)-1-carboxyethyl-2-
azetidinone (5). T. Kawabata, Y. Kimura, Y. Ito, S. Terashima, A. Sasaki, and M$ T. Kawabata, Y. Kimura, Y. Ito, S. Terashima, A. Sasaki, and M. Sunagawa, *Tetmhedmn L&t., 27,6241(1986). Zdem, Tetrahedron, 44,2149* **(1988).**
- **The** major byproducts of the [2+21-cycloaddition reaction were found to be N-(di-p-anisylmethyl)acetoacetamide (<20%) and dehydroacetic acid. The latter acid could be easily removed by extracting the reaction mixture with 2M NaOH.
- 22) The racemic β -lactam (dl-13a) was prepared from dl-9a according to the same synthetic route as described for the preparation of optically active 13a (Y. Ito and S. Terashima, unpublished results).
- K G. Holden, *"Chemistry and Biology of &Lactam Antibiotics",* Vol. 2, ed. R. B. Morin and M. Gorman, Academic Press, Inc., New York-London, p. 114 (1982).
- F. A. Bouffard and B. G. Christensen, J. Org. *Chem., 46,2208* (1981).
- 25) Y. Ito, T. Katsuki, and M. Yamaguchi, *Tetrahedron Lett.*, 26, 4643 (1985).
- 0. Mitsunobu, *Synthesis,* **l(1981).**
- G. Guanti, L. Banfi, E. Narisano, C. Scolastico, and E. Bosone, *Synthesis, 609* **(1985).**
- **After the** preliminary communications had been published (see reference 11, Cainelli *et al.* reported direct cleavage of the 1,2-amido alcohol with lead tetraacetate in the presence of calcium carbonate to afford the 4-acetoxy-2-azetidinone derivative. G. Cainelli, M. calcium carbonate to afford the 4-acetoxy-2-azetidinone derivative. Panunzio, D. Giacomini, G. Martelli, and G. Spunta, *J. Am. Chem. Soc.*, 110, 6879 (1988).
- **29)** E. J. Corey and C. U. Kim, J. **Am.** *Chem. Sot., 94,* **7586 (1972).** *Zdem, Tetrahedron L&t., 287* **(1974).** *Org. Syn., 58,122* (1978).
- 30) Although other oxidants such as potassium permanganate, ruthenium(II1) chloridesodium periodate, ruthenium(II1) chloride-sodium bromate, sulfur trioxide pyridine complex-dimethylsulfoxide-triethylamine, acetic anhydride-dimethylsulfoxide-triethylamine, and chlorine-dimethylsulfide-triethylamine were also examined for oxidizing 22 to 23, satisfactory results could not be obtained due to N -chlorination, formation of the methylthiomethyl ether, and/or recovery of the starting material.
- **31) P. G. M. Wuts and S. S. Bigelow,** *J. Org. Chem., 48,3489* **(1983).**
- 32) P. H. J. Carlsen, T. Katsuki, and K. B. Sharpless, *J. Org. Chem.*, **46**, 3936 (1981).